

Maxwell's Equations and QED: Which Is Fact and Which Is Fiction?

Randell L. Mills



Abstract

The claim that quantum electrodynamics (QED) is the most successful theory in history is critically evaluated. The Dirac equation was postulated in 1926 as a means to remedy the nonrelativistic nature of the Schrödinger equation to provide the missing fourth quantum number. The positive and negative square root terms provided an argument for the existence of negative energy states of the vacuum, virtual particles, and corresponding so-called QED computer algorithms for calculating unexpected observables such as the Lamb shift and the anomalous magnetic moment of the electron. Dirac's original attempt to solve the bound electron physically with stability with respect to radiation according to Maxwell's equations, with the further constraints that it be relativistically invariant and give rise to electron spin, is achievable using a classical approach. Starting with the same essential physics as Bohr, Schrödinger, and Dirac of e^- moving in the Coulombic field of the proton and the wave equation as an equation of motion rather than energy after Schrödinger, advancements in the understanding of the stability of the bound electron to radiation are applied to solve for the exact nature of the electron. Rather than using the postulated Schrödinger boundary condition " $\Psi \rightarrow 0$ as $r \rightarrow \infty$," which leads to a purely mathematical model of the electron, the constraint is based on experimental observation. Using Maxwell's equations, the classical wave equation is solved with the constraint that the bound ($n = 1$)-state electron cannot radiate energy. Although it is well known that an accelerated point particle radiates, an extended distribution modeled as a superposition of accelerating charges does not have to radiate. A simple invariant physical model arises naturally wherein the predicted results are extremely straightforward and internally consistent, requiring minimal mathematics, as in the case of the most famous equations of Newton, Maxwell, Lorentz, de Broglie, and Planck on which the model is based. No new physics is needed; only the known physical laws based on direct observation are used. Rather than invoking untestable "flights of fancy," the results of QED, such as the anomalous magnetic moment of the electron, the Lamb shift, the fine structure and hyperfine structure of the hydrogen atom, and the hyperfine structure intervals of positronium and muonium, can be solved exactly from Maxwell's equations to the limit possible based on experimental measurements, which confirms QED's illegitimacy as representative of reality.


Key words: QED, Maxwell's equations, Lamb shift, fine structure and hyperfine structure of the hydrogen atom, hyperfine structure intervals of positronium and muonium

1. INTRODUCTION

The hydrogen atom is the only real problem for which the Schrödinger equation can be solved without approximations; however, it only provides three quantum numbers — not four. Furthermore, the Schrödinger equation is not accurate at all. It is nonrelativistic, and

there are major differences between predicted and experimental ionization energies as Z increases, and inescapable disagreements between observation and predictions arise from the later-postulated Dirac equation as well as the Schrödinger equation.^(1–10) In addition to spin, it misses the Lamb shift, the fine structure,

and the hyperfine structure completely; it is not stable to radiation; and it has many other problems with predictions that do not match experimentation.^(2–10) It also has an infinite number of solutions, not just the ones given in textbooks, as given in Margenau and Murphy⁽¹¹⁾ and Ref. 9.

Unlike physical laws such as Maxwell's equations, it is always disconcerting to those that study quantum mechanics (QM) that both QM and quantum electrodynamics (QED) must be accepted without any underlying physical s for fundamental observables such as the stability of the hydrogen atom in the first place. In this instance a circular argument regarding definitions for parameters in the wave equation solutions and the Rydberg series of spectral lines replaces a first principles–based prediction of those lines.^(2–10)

Nevertheless, it is felt that applying the Schrödinger equation to real problems has provided useful approximations for physicists and chemists. Schrödinger interpreted $e\Psi^*(x)\Psi(x)$ as the charge-density or the amount of charge between x and $x + dx$ (Ψ^* is the complex conjugate of Ψ). Presumably, then, he pictured the electron to be spread over large regions of space. After Schrödinger's interpretation, Max Born, who was working with scattering theory, found that this interpretation led to inconsistencies, and he replaced the Schrödinger interpretation with the probability of finding the electron between x and $x + dx$ as

$$\int \Psi(x)\Psi^*(x)dx. \quad (1)$$

Born's interpretation is generally accepted. Nonetheless, interpretation of the wave-function is a never-ending source of confusion and conflict. Many scientists have solved this problem by conveniently adopting the Schrödinger interpretation for some problems and the Born interpretation for others. This duality allows the electron to be everywhere at one time — yet have no volume. Alternatively, the electron can be viewed as a discrete particle that moves here and there (from $r = 0$ to $r = \infty$), and $\Psi\Psi^*$ gives the time average of this motion.

Despite its successes, QM has remained mysterious to all who have encountered it. Starting with Bohr and progressing into the present, the departure from intuitive, physical reality has widened. The connection between QM and reality is more than just a “philosophical” issue. It reveals that QM is not a correct or complete theory of the physical world and that inescapable internal inconsistencies and incongruities arise when attempts are made to treat it as a

physical as opposed to a purely mathematical “tool.” Some of these issues are discussed in a review by Laloë.⁽¹⁾

But QM has severe limitations even as a tool. Beyond one-electron atoms, multielectron-atom quantum-mechanical equations cannot be solved except by approximation methods⁽¹²⁾ involving adjustable-parameter theories (perturbation theory, variational methods, self-consistent field method, multiconfiguration Hartree–Fock method, multiconfiguration parametric potential method, $1/Z$ expansion method, multiconfiguration Dirac–Fock method, electron correlation terms, QED terms, etc.) — all of which contain assumptions that cannot be physically tested and are not consistent with physical laws. And calling the substitutes approximations is misleading. They are not approximations since they involve new physics or constructs or are simply curve-fitting algorithms, as discussed previously.⁽⁶⁾ With adjustable-parameter methods, it is necessary to repeat trial-and-error experimentation to find which method of calculation gives the right answer. It is common practice to present only the successful procedure as if it followed from first principles and not mention the actual method by which it was found. In many cases the success of QM can be attributed to the use of arbitrary variational parameters in all-space probability wavefunctions and arbitrary renormalization of intrinsic infinities in the corresponding energies. Furthermore, the distinction between series expansion or variation of a physical parameter of an equation based on a physical action and the fabrication of actions based on untestable constructs corresponding to a series with variational (adjustable) parameters is discussed in Section 2.

Also, after decades of futility, QM and the intrinsic Heisenberg uncertainty principle have not yielded a unified theory, are still purely mathematical, and have yet to be shown to be based in reality.^(2,7,9) Both are based on circular arguments that the electron is a point with no volume with a vague probability wave requiring that the electron have infinite numbers of positions and energies, including negative and infinite energies simultaneously. It may be time to revisit the 75-year-old notion that fundamental particles such as the electron are one- or zero-dimensional and obey different physical laws than objects comprising fundamental particles and the even more disturbing view that fundamental particles don't obey physical laws — rather they obey mathematics devoid of physical laws. Perhaps mathematics does not determine physics. It only models physics.

The Schrödinger equation was originally postulated in 1926 as having a solution of the one-electron atom. It gives the principal energy levels of the hydrogen atom as eigenvalues of eigenfunction solutions of the Laguerre differential equation. But, as the principal quantum number $n \gg 1$, the eigenfunctions become nonsensical since they are sinusoidal over all space; thus they are nonintegrable, cannot be normalized, and are infinite.⁽¹³⁾ Despite its wide acceptance, on deeper inspection, the Schrödinger equation solution is plagued with many failings, as well as difficulties in terms of a physical interpretation, that have caused it to remain controversial since its inception. Only the one-electron atom may be solved without approximations, the results are very poor, and it fails to predict electron spin and leads to models with nonsensical consequences such as negative energy states of the vacuum, infinities, and negative kinetic energy. In addition to many predictions that simply do not agree with observations, the Schrödinger equation and succeeding extensions predict noncausality, nonlocality, spooky actions at a distance or quantum telepathy, perpetual motion, and many internal inconsistencies where contradicting statements have to be taken true simultaneously.^(2,7,9)

It was reported previously⁽⁹⁾ that the behavior of free electrons in superfluid helium has again forced the issue of the meaning of the wave-function. Electrons form bubbles in superfluid helium that reveal that the electron is real and that a physical interpretation of the wave-function is necessary. Furthermore, when irradiated with light of energy of about 0.5 to several electron volts,^(9,14) the electrons carry current at different rates as if they exist with different sizes. The nature of the wave-function needs to be addressed. It is time for the physical rather than the mathematical nature of the wave-function to be determined. A classical derivation based on an extended electron was shown previously to be in complete agreement with observations, whereas QM has no utility.^(7,9)

From the time of its inception, QM has been controversial because its foundations are in conflict with physical laws and are internally inconsistent. Interpretations of QM such as hidden variables, multiple worlds, consistency rules, and spontaneous collapse have been put forward in an attempt to base the theory in reality. Unfortunately, many theoreticians ignore the requirement that the wave-function be real and physical in order for it to be considered a valid description of reality. For example, regarding this issue, Fuchs and Peres believe,⁽¹⁵⁾ “Contrary to those

desires, quantum theory does *not* describe physical reality. What it does is provide an algorithm for computing *probabilities* for macroscopic events (‘detector ticks’) that are the consequences of our experimental interventions. This strict definition of the scope of quantum theory is the only interpretation ever needed, whether by experimenters or theorists.”

With Penning traps, it is possible to measure transitions including those with hyperfine levels of electrons of single ions. This case can be experimentally distinguished from statistics over equivalent transitions in many ions. Whether many or one, the transition energies are always identical within the resonant line width. So *probabilities* have no place in describing atomic energy levels. Moreover, quantum theory is incompatible with probability theory since it is based on underlying unknown, but determined, outcomes, as discussed previously.⁽⁹⁾

Wave-function solutions of the Schrödinger equation are interpreted as probability-density functions. Quantum theory confuses the concepts of a wave and a probability-density function, which are based on totally different mathematical and physical principles. The use of “probability” in this instance does not conform to the mathematical rules and principles of probability theory. Statistical theory is based on an existing deterministic reality with incomplete information, whereas quantum measurement acts on a “probability-density function” to determine a reality that did not exist before the measurement. Additionally, it is nonsensical to treat a single particle such as an electron as if it were a population of electrons and to assign the single electron to a statistical distribution over many states. The electron has conjugate degrees of freedom such as position, momentum, and energy that obey conservation laws in an inverse- r Coulomb field. A single electron cannot have multiple positions and momenta or energies simultaneously.

The Copenhagen interpretation provides another meaning of QM. It asserts that what we observe is all we can know; any speculation about what an electron, photon, atom, or other atomic-sized entity is really or what it is doing when we are not looking is just that — speculation. The postulate of quantum measurement asserts that the process of measuring an observable forces it into a state of reality. In other words, reality is irrelevant until a measurement is made. In the case of electrons in superfluid helium, the fallacy with this position is that the “ticks” (migration times of electron bubbles) reveal that the electron is real before a measurement is made. Furthermore, experiments on transitions on single ions such as Ba^+ in a

Penning trap under continuous observation demonstrate that the postulate of quantum measurement of QM is experimentally disproved, as discussed previously.^(9,16) These issues and other such flawed philosophies and interpretations of experiments that arise from QM were discussed previously.⁽¹⁻¹⁰⁾

QM gives correlations with experimental data. It does not explain the mechanism for the observed data. But it should not be surprising that it may give good correlations given that the constraints of internal consistency and conformance to physical laws are removed for a wave equation with an infinite number of solutions wherein the solutions may be formulated as an infinite series of eigenfunctions with variable parameters. There are no physical constraints on the parameters. They may even correspond to unobservables such as virtual particles, hyperdimensions, effective nuclear charge, polarization of the vacuum, worm holes, spooky action at a distance, infinities, parallel universes, faster than light travel, etc. If you invoke the constraints of internal consistency and conformance to physical laws, QM has never successfully solved a physical problem, as discussed previously.⁽⁶⁾

Reanalysis of old experiments and many new experiments, including electrons in superfluid helium, challenge the Schrödinger equation predictions. Many noted physicists rejected QM. Feynman also attempted to use first principles, including Maxwell's equations, to discover new physics to replace QM.⁽¹⁷⁾ Other great physicists of the 20th century searched. "Einstein [...] insisted [...] that a more detailed, wholly deterministic theory must underlie the vagaries of quantum mechanics."⁽¹⁸⁾ He felt that scientists were misinterpreting the data. These issues and the results of many experiments, such as the wave-particle duality, the Lamb shift, anomalous magnetic moment of the electron, and transition and decay lifetimes; experiments invoking interpretations of spooky action at a distance such as the Aspect experiment; entanglement; and double slit-type experiments, are shown to be absolutely predictable and physical in the context of a theory of classical quantum mechanics (CQM) derived from first principles.⁽²⁻¹⁰⁾

2. QED

QM failed to predict the results of the Stern-Gerlach experiment, which indicated the need for an additional quantum number. In QM the spin angular momentum of the electron is called the "intrinsic angular momentum" since no physical interpretation

exists. (Currents corresponding to the observed magnetic field of the electron cannot exist in one dimension of four-dimensional space-time, where Ampère's law and the intrinsic special relativity determine the corresponding unique current.) The Schrödinger equation is not Lorentzian invariant, in violation of special relativity. The Schrödinger equation also misses the Lamb shift, the fine structure, and the hyperfine structure completely, and it is not stable to radiation. QED was proposed by Dirac in 1926 to provide a generalization of QM for high energies in conformity with the theory of special relativity and to provide a consistent treatment of the interaction of matter with radiation. But it does not bridge the gap between QM and special relativity. From Weisskopf,⁽¹⁹⁾ "Dirac's quantum electrodynamics gave a more consistent derivation of the results of the correspondence principle, but it also brought about a number of new and serious difficulties." QED (1) does not explain nonradiation of bound electrons, (2) contains an internal inconsistency with special relativity regarding the classical electron radius — the electron mass corresponding to its electric energy is infinite, (3) admits solutions of negative rest mass and negative kinetic energy, (4) leads to infinite kinetic energy and infinite electron mass for the interaction of the electron with the predicted zero-point field fluctuations, and (5) still yielded infinities when Dirac used the unacceptable states of negative mass for the description of the vacuum. Dirac's postulated relativistic wave equation gives the inescapable result of a cosmological constant that is at least 120 orders of magnitude larger than the best observational limit, due to the unacceptable states of negative mass for the description of the vacuum, as discussed previously.^{1,(2-7,9,10)} The negative mass states further create an absolute "ether"-like frame, in violation of special relativity, which was disproved by the Michelson-Morley experiment.

In retrospect, Dirac's equation, which was postulated to explain spin, relies on the unfounded notions of negative energy states of the vacuum, virtual particles, and gamma factors; thus it cannot be the correct description of a bound electron even though it gives an additional quantum number interpreted as corresponding to the phenomenon of electron spin. Ironically, it is not even internally consistent with respect to its intent of being in accord with special relativity. The Dirac equation violates Maxwell's equations with respect to stability to radiation; contains an internal inconsistency with special relativity regarding the classical electron radius and

states of negative rest mass and negative kinetic energy, as given by Weisskopf,⁽¹⁹⁾ and further violates Einstein causality and locality in addition to conservation of energy, as shown by the Klein paradox discussed previously.^{2,(2,4,7)} Furthermore, everyday observation demonstrates that causality and locality always hold. Einstein also argued that a probabilistic versus deterministic nature of atomic particles leads to disagreement with special relativity. In fact, the nonlocality result of the Copenhagen interpretation violates causality, as shown by Einstein, Podolsky, and Rosen (EPR) in a classic paper⁽²²⁾ that presented a paradox involving instantaneous (faster-than-light) communication between particles called “spooky action at a distance,” which led them to conclude that QM is not a complete or correct theory. The implications of the EPR paper and the exact Maxwellian predictions of “spooky action” and “entanglement” experiments, incorrectly interpreted in the context of QM, are given in Chapter 42 of Ref. 7.

In 1947, contrary to Dirac’s predictions, Lamb discovered a 1000 MHz shift between the $^2S_{1/2}$ state and the $^2P_{1/2}$ state of the hydrogen atom.⁽²⁴⁾ This so-called Lamb shift marked the beginning of modern QED. In the words of Dirac,⁽²⁵⁾ “No progress was made for 20 years. Then a development came initiated by Lamb’s discovery and explanation of the Lamb Shift, which fundamentally changed the character of theoretical physics. It involved setting up rules for discarding ... infinities....” Renormalization is currently believed to be required of any fundamental theory of physics.⁽²⁶⁾ However, dissatisfaction with renormalization has been expressed at various times by many physicists, including Dirac,⁽²⁷⁾ who felt that, “This is just not sensible mathematics. Sensible mathematics involves neglecting a quantity when it turns out to be small — not neglecting it just because it is infinitely great and you do not want it!”

Although the Dirac equation did not predict the Lamb shift or the electron g factor,^(24,28,29) its feature of negative-mass states of the vacuum gave rise to the postulates of QED that have become a centerpiece of QM to explain these and other similar observations. One of QED’s seminal aspects of renormalization, which was subsequently grafted onto atomic theory, was a turning point in physics similar to the decision to treat the electron as a point particle/probability wave, a point with no volume with a vague probability wave requiring that the electron have an infinite number of positions and energies, including negative and infinite energies simultaneously. The adoption of

the probabilistic versus deterministic nature of atomic particles violates all physical laws, including special relativity with violation of causality, as pointed out by Einstein⁽²²⁾ and de Broglie.⁽³⁰⁾ Consequently, it was rejected even by Schrödinger.⁽³¹⁾

Pure mathematics took the place of physics when calculating subtle shifts of the hydrogen atomic energy levels. Moreover, in QED, the pure mathematics approach has been confused with physics to the point that virtual particles are really considered as causing the observable. The justification for the linkage is often incorrectly associated with the usage of series expansion and variational methods to solve problems based on physical laws. But series expansion of an equation based on a physical action or variation of a physical parameter of the equation versus the fabrication of an action based on fantastical untestable constructs that are represented by a series are clearly different. For example, the motion of a pendulum can be solved exactly in terms of an elliptic integral using Newtonian mechanics. Expanding the elliptic integral in a power series and ignoring negligible terms in the series and setting up arbitrary rules for *discarding infinities* are clearly not the same. Furthermore, inventing virtual particles that have an action on space, and subsequently on an electron, and expanding terms in the energy equation due to a gravitating body causing a gravitational field and thus an action on the pendulum are very different. In QED, virtual particles are not merely a substitutional or expansion variable. They are really considered as causing the observable.

In a further exercise of poor science, virtual particle-based calculations are even included in the determination of the fundamental constants, which are circularly used to calculate the parameter ascribed to the virtual particles. For example, using the electron magnetic moment anomaly in the selection of the best value of the fine-structure constant, the CODATA publication⁽³²⁾ reports the use of virtual particles:

The term A_1 is mass independent and the other terms are functions of the indicated mass ratios. For these terms the lepton in the numerator of the mass ratio is the particle under consideration, while the lepton in the denominator of the ratio is the virtual particle that is the source of vacuum polarization that gives rise to the term.

There is no direct evidence that virtual particles exist or that they polarize the vacuum. Even their postulation is an oxymoron.

Throughout the history of quantum theory, whenever there was an advance to a new application, it was necessary to repeat a trial-and-error experimentation to find which method of calculation gave the right answers. Often the textbooks present only the successful procedure as if it followed from first principles and do not mention the actual method by which it was found. In electromagnetic theory based on Maxwell's equations one deduces the computational algorithm from the general principles. In quantum theory the logic is just the opposite. One chooses the principle (e.g., phenomenological Hamiltonians) to fit the empirically successful algorithm. For example, we know that it required a great deal of art and tact over decades of effort to get correct predictions out of QED. The QED method of determining $(g - 2)/2$ from the *postulated* Dirac equation is based on a *postulated* power series of α/π , where each *postulated* virtual particle is a source of *postulated* vacuum polarization that gives rise to a *postulated* term, which is processed over decades using ad hoc rules to remove infinities from each term that arises from *postulated* scores of *postulated* Feynman diagrams. The solution so obtained using the perturbation series further requires a *postulated* truncation since the series *diverges*. Mohr and Taylor reference some of the Herculean efforts to arrive at g using QED:⁽³²⁾

the sixth-order coefficient $A_1^{(6)}$ arises from 72 diagrams and is also known analytically after nearly 30 years of effort by many researchers (see Roskies, Remiddi, and Levine (1990) for a review of the early work). It was not until 1996 that the last remaining distinct diagrams were calculated analytically, thereby completing the theoretical expression for $A_1^{(6)}$.

For the right experimental numbers to emerge one must do the calculation (i.e., subtract off the infinities) in one particular way and not in some other way that appears in principle equally valid. For example, Milonni⁽³³⁾ presents a QED derivation of the magnetic moment of the electron that gives a result of the wrong sign and requires the introduction of an

upper limit K in the integration over $k = \omega/c$ in order to avoid a divergence.

A differential mass is arbitrarily added, and then

the choice $K = 0.42mc/\hbar$ yields $(g - 2)/2 = \alpha/2\pi$ which is the relativistic QED result to first order

in α [...] However, the reader is warned not to take these calculations too seriously, for the result $(g - 2)/2 = \alpha/2\pi$ could be obtained by retaining only the first (radiation reaction) term in (3.112) and choosing $K = 3mc/8\hbar$. It should also be noted that the solution $K \cong 0.42mc/\hbar$ of (3.112) with $(g - 2)/2 = \alpha/2\pi$ is not unique.

Such an ad hoc nonphysical approach makes incredible

the cliché that QED is the best theory we have!⁽³⁴⁾

or the statement that

The history of quantum electrodynamics (QED) has been one of unblemished triumph.⁽³⁵⁾

There is a corollary, noted by Kallen: from an inconsistent theory, any result may be derived.

In an attempt to provide some physical insight into atomic problems and starting with the same essential physics as Bohr of e^- moving in the Coulombic field of the proton and the wave equation as modified after Schrödinger, a classical approach was explored that yields a remarkably accurate model and provides insight into physics on the atomic level.⁽²⁻⁷⁾ Physical laws and intuition are restored when dealing with the wave equation and quantum-mechanical problems. Specifically, a theory of CQM was derived from first principles that successfully applies physical laws on all scales. Rather than using the *postulated* Schrödinger boundary condition " $\Psi \rightarrow 0$ as $r \rightarrow \infty$," which leads to a purely mathematical model of the electron, the constraint is based on experimental observation. Using Maxwell's equations, *the classical wave equation as an equation of motion is solved with the constraint that the bound ($n = 1$)-state electron cannot radiate energy.* The electron must be extended, rather than a point. On this basis, with the assumption that physical laws including Maxwell's equation apply to bound electrons, the hydrogen atom was solved exactly from first principles. The remarkable agreement across the spectrum of experimental results indicates that this is the correct model of the hydrogen atom.

It was shown previously that QM does not explain the stability of the atom to radiation,⁽²⁾ whereas the Maxwellian approach gives a natural relationship between Maxwell's equations, special relativity, and general relativity. CQM holds over a scale of space-time of 85 orders of magnitude — it correctly predicts

the nature of the universe from the scale of the quarks to that of the cosmos.⁽³⁾ A review is given by Landvogt.⁽³⁶⁾ In a third paper the atomic physical approach was applied to multielectron atoms that were solved exactly, disproving the deep-seated view that such exact solutions cannot exist according to QM. The general solutions for one- through twenty-electron atoms are given in Ref. 4. The predictions are in remarkable agreement with the experimental values known for 400 atoms and ions. A fourth paper presents a solution based on physical laws and fully compliant with Maxwell's equations that solves the 26 parameters of molecular ions and molecules of hydrogen isotopes in closed-form equations with fundamental constants only that match the experimental values.⁽⁵⁾ In a fifth paper the nature of atomic physics being correctly represented by QM versus CQM is subjected to a test of internal consistency for the ability to calculate the conjugate observables using the same solution for each of the separate experimental measurements.⁽⁶⁾ It is confirmed that the CQM solution is the accurate model of the helium atom by the agreement of predicted and observed conjugate parameters of the free electron, ionization energy of helium and all two-electron atoms, ionization energies of multielectron atoms, electron scattering of helium for all angles, and all He I excited states, using the same unique physical model in all cases. Over 500 conjugate parameters are calculated using a unique solution of the two-electron atom without any adjustable parameters. In the closed-form equations overall agreement is achieved to the level obtainable considering the error in the measurements and in the fundamental constants.

In contrast, the quantum theory fails utterly. Ad hoc computer algorithms are used to generate meaningless numbers with internally inconsistent and nonphysical models that have no relationship to physics. Attempts are often made to numerically reproduce prior theoretical numbers using adjustable parameters, including arbitrary wave-functions in computer programs with precision that is often much greater (e.g., eight significant figures greater) than possible based on the propagation of errors in the measured fundamental constants implicit in the physical problem.

In this sixth paper of a series, rather than invoking renormalization, untestable virtual particles, and polarization of the vacuum by the virtual particles, the results of QED such as the anomalous magnetic moment of the electron, the Lamb shift, the fine structure and hyperfine structure of the hydrogen atom, and the hyperfine structure intervals of posi-

tronium and muonium (thought to be only solvable using QED) are solved exactly from Maxwell's equations to the limit possible based on experimental measurements.

3. CLASSICAL QUANTUM THEORY OF THE ATOM BASED ON MAXWELL'S EQUATIONS

In this paper the old view that the electron is a zero- or one-dimensional point in an all-space probability wave-function $\Psi(x)$ is not taken for granted. The theory of CQM, derived from first principles, must successfully and consistently apply physical laws on all scales.⁽²⁻¹⁰⁾ Stability to radiation was ignored by all past atomic models. Historically, the point at which QM broke with classical laws can be traced to the issue of nonradiation of the one-electron atom. Bohr just postulated orbits stable to radiation with the further postulate that the bound electron of the hydrogen atom does not obey Maxwell's equations — rather it obeys different physics.^(2,7) Later physics was replaced by “pure mathematics” based on the notion of the inexplicable wave-particle duality nature of electrons, which led to the Schrödinger equation, wherein the consequences of radiation predicted by Maxwell's equations were ignored. Ironically, Bohr, Schrödinger, and Dirac used the Coulomb potential, and Dirac used the vector potential of Maxwell's equations. But all ignored electrodynamics and the corresponding radiative consequences. Dirac originally attempted to solve the bound electron physically with stability with respect to radiation according to Maxwell's equations with the further constraints that it be relativistically invariant and give rise to electron spin.⁽³⁷⁾ He and many founders of QM, such as Sommerfeld, Bohm, and Weinstein, wrongly pursued a planetary model, were unsuccessful, and resorted to the current mathematical probability-wave model that has many problems.^(1-10,19,22,23,37) Consequently, Feynman, for example, attempted to use first principles, including Maxwell's equations, to discover new physics to replace QM.⁽³⁸⁾

Physical laws may indeed be the root of the observations thought to be “purely quantum mechanical,” and it may have been a mistake to assume that Maxwell's electrodynamic equations must be rejected at the atomic level. Thus, in the present approach, the classical wave equation is solved with the constraint that a bound ($n = 1$)-state electron cannot radiate energy.

Herein, derivations consider the electrodynamic effects of moving charges as well as the Coulomb

potential, and the search is for a solution representative of the electron wherein there is acceleration of charge motion without radiation. The mathematical formulation for zero radiation based on Maxwell's equations follows from a derivation by Haus.⁽³⁹⁾ The function that describes the motion of the electron must not possess space-time Fourier components that are synchronous with waves traveling at the speed of light. Similarly, nonradiation is demonstrated based on the electron's electromagnetic fields and the Poynting power vector.

It was shown previously⁽³⁻⁸⁾ that CQM gives closed-form solutions for the atom, including the stability of the $n = 1$ state and the instability of the excited states, the equation of the photon and electron in excited states, and the equation of the free electron and photon, which predict the wave-particle duality behavior of particles and light. The current- and charge-density functions of the electron may be directly physically interpreted. For example, spin angular momentum results from the motion of negatively charged mass moving systematically, and the equation for angular momentum, $\mathbf{r} \times \mathbf{p}$, can be applied directly to the wave-function (a current-density function) that describes the electron. The magnetic moment of a Bohr magneton, Stern–Gerlach experiment, g factor, Lamb shift, resonant line width and shape, selection rules, correspondence principle, wave-particle duality, excited states, state lifetimes, reduced mass, rotational energies, momenta, orbital and spin splitting, spin-orbit coupling, Knight shift, spin-nuclear coupling, and elastic electron scattering from helium atoms are derived in closed-form equations based on Maxwell's equations. The calculations agree with experimental observations.

In contrast to the failure of the Bohr theory and the nonphysical, adjustable-parameter approach of QM, multielectron atoms^(4,7) and the nature of the chemical bond^(5,7) are given by exact closed-form solutions containing fundamental constants only. Using the nonradiative wave equation solutions that describe each bound electron having conserved momentum and energy, the radii are determined from the force balance of the electric, magnetic, and centrifugal forces that correspond to the minimum of energy of the atomic or ionic system. The ionization energies are then given by the electric and magnetic energies at these radii. The spreadsheets to calculate the energies from exact solutions of one- through twenty-electron atoms are available from the Internet.⁽⁴⁰⁾ For 400 atoms and ions the agreement between the predicted and experimental results is remarkable.

3.1 One-Electron Atoms

One-electron atoms include the hydrogen atom, He^+ , Li^{2+} , Be^{3+} , and so on. The mass-energy and angular momentum of the electron are constant; this requires that the equation of motion of the electron be temporally and spatially harmonic. Thus the classical wave equation applies and

$$\left[\nabla^2 - \frac{1}{v^2} \frac{\partial^2}{\partial t^2} \right] \rho(r, \theta, \phi, t) = 0, \quad (2)$$

where $\rho(r, \theta, \phi, t)$ is the time-dependent charge-density function of the electron in time and space. In general, the wave equation has an infinite number of solutions. To arrive at the solution that represents the electron, a suitable boundary condition must be imposed. It is well known from experiments that each single atomic electron of a given isotope radiates to the same stable state. Thus the physical boundary condition of nonradiation of the bound electron was imposed on the solution of the wave equation for the time-dependent charge-density function of the electron.⁽²⁻⁸⁾ The condition for radiation by a moving point charge given by Haus⁽³⁹⁾ is that its space-time Fourier transform possess components that are synchronous with waves traveling at the speed of light. Conversely, it is proposed that the condition for nonradiation by an ensemble of moving charge that makes up a current-density function is as follows:

For nonradiative states the current-density function must not possess space-time Fourier components that are synchronous with waves traveling at the speed of light.

The time, radial, and angular solutions of the wave equation are separable. The motion is time harmonic with frequency ω_n . A constant angular function is a solution to the wave equation. Solutions of the Schrödinger wave equation comprising a radial function radiate according to Maxwell's equation, as shown previously by application of Haus's condition.⁽³⁹⁾ In fact, it was found that any function that permitted radial motion gave rise to radiation. A radial function that does satisfy the boundary condition is a radial delta function

$$f(r) = \frac{1}{r^2} \delta(r - r_n). \quad (3)$$

This function defines a constant charge density on a spherical shell, where $r_n = nr_1$, wherein n is an integer in an excited state, and (2) becomes the two-dimensional wave equation plus time with separable time and angular functions. Given time-harmonic motion and a radial delta function, the relationship between an allowed radius and the electron wavelength is given by

$$2\pi r_n = \lambda_n, \quad (4)$$

where the integer subscript n here and in (3) is determined during photon absorption as given in the Excited States of the One-Electron Atom (Quantization) section of Ref. 7. Using the observed de Broglie relationship for the electron mass where the coordinates are spherical,

$$\lambda_n = \frac{h}{p_n} = \frac{h}{m_e v_n}, \quad (5)$$

and the magnitude of the velocity for *every* point on the electron surface is

$$v_n = \frac{\hbar}{m_e r_n}. \quad (6)$$

The sum of the $|\mathbf{L}_i|$, the magnitude of the angular momentum of each infinitesimal density element of the element of mass m_i , must be constant. The constant is \hbar :

$$\sum |\mathbf{L}_i| = \sum |\mathbf{r} \times m_i \mathbf{v}| = m_e r_n \frac{\hbar}{m_e r_n} = \hbar. \quad (7)$$

Thus an electron is a two-dimensional spherical current surface (zero thickness), called an *electron orbitsphere*, shown in Fig. 1, that can exist in a bound state at only specified distances from the nucleus determined by an energy minimum. The corresponding current, which gives rise to the phenomenon of *spin*, is derived in Section 3.2 using the current vector field shown in Fig. 2. (See the Orbitsphere Equation of Motion for $\ell = 0$ of Ref. 7 at Chapter 1.)

Nonconstant functions are also solutions for the angular functions. To be a harmonic solution of the wave equation in spherical coordinates, these angular functions must be spherically harmonic functions.⁽⁴¹⁾ A zero of the space-time Fourier transform of the product

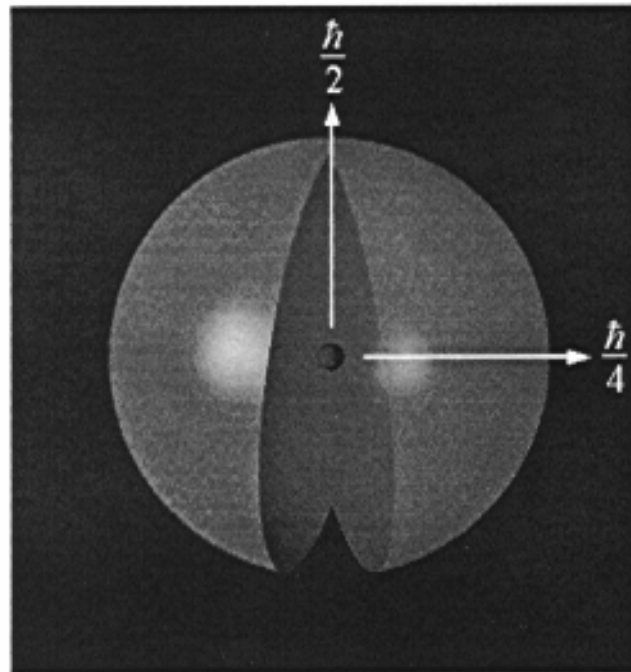


Figure 1. The orbitsphere is a two-dimensional spherical shell of current having zero thickness with the Bohr radius of the hydrogen atom, $r = a_H$. The current and charge densities confined to two dimensions at $r_n = nr_1$ are uniform.

function of two spherically harmonic angular functions, a time-harmonic function, and an unknown radial function is sought. The solution for the radial function that satisfies the boundary condition is also a delta function given by (3). Thus bound electrons are described by a charge-density (mass-density) function that is the product of a radial delta function, two angular functions (spherically harmonic functions), and a time-harmonic function:

$$\rho(r, \theta, \phi, t) = f(r)A(\theta, \phi, t) = \frac{1}{r^2} \delta(r - r_n)A(\theta, \phi, t), \quad (8)$$

$$A(\theta, \phi, t) = Y(\theta, \phi)k(t).$$

In these cases the spherically harmonic functions correspond to a traveling charge-density wave confined to the spherical shell, which gives rise to the phenomenon of orbital angular momentum. The orbital functions that modulate the constant “spin” function shown graphically in Fig. 3 are given in Section 3.3.

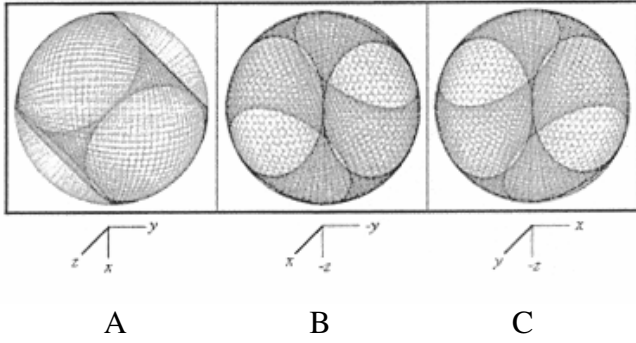


Figure 2A–C. The orbitosphere current vector field pattern from the perspective of looking along the z axis, x axis, and y axis, respectively.

3.2 Spin Function

The orbitosphere spin function comprises a constant charge- (current-) density function with moving charge confined to a two-dimensional spherical shell. The magnetostatic current pattern of the orbitosphere spin function comprises an infinite series of correlated orthogonal great circle current loops wherein each charge- (current-) density element moves time harmonically with constant angular velocity

$$\omega_n = \frac{\hbar}{m_e r_n^2}. \quad (9)$$

The uniform current density function $Y_0^0(\theta, \phi)$ that gives rise to the spin of the electron is generated from a basis set current-vector field defined as the orbitosphere current-vector field (“orbitosphere-cvf”). The orbitosphere-cvf comprises a continuum of correlated orthogonal great circle current loops. The current pattern comprising two components is generated over the surface by two sets (*steps 1 and 2*) of rotations of two orthogonal great circle current loops that serve as basis elements about each of the $(\mathbf{i}_x, \mathbf{i}_y, 0\mathbf{i}_z)$ and $(-\mathbf{i}_x/\sqrt{2}, \mathbf{i}_y/\sqrt{2}, \mathbf{i}_z)$ axes, respectively, by π radians. In Appendix III of Ref. 7 the *continuous* uniform electron current density function $Y_0^0(\theta, \phi)$ with the same angular momentum components, $\mathbf{L}_z = \hbar/2$ and $\mathbf{L}_{xy} = \hbar/4$, is then exactly generated from this orbitosphere-cvf as a basis element by a convolution operator comprising an autocorrelation-type function.

The orthogonal great circle basis set for step 1 is shown in Fig. 4. One half of the orbitosphere-cvf, the orbitosphere-cvf component of step 1, is generated by

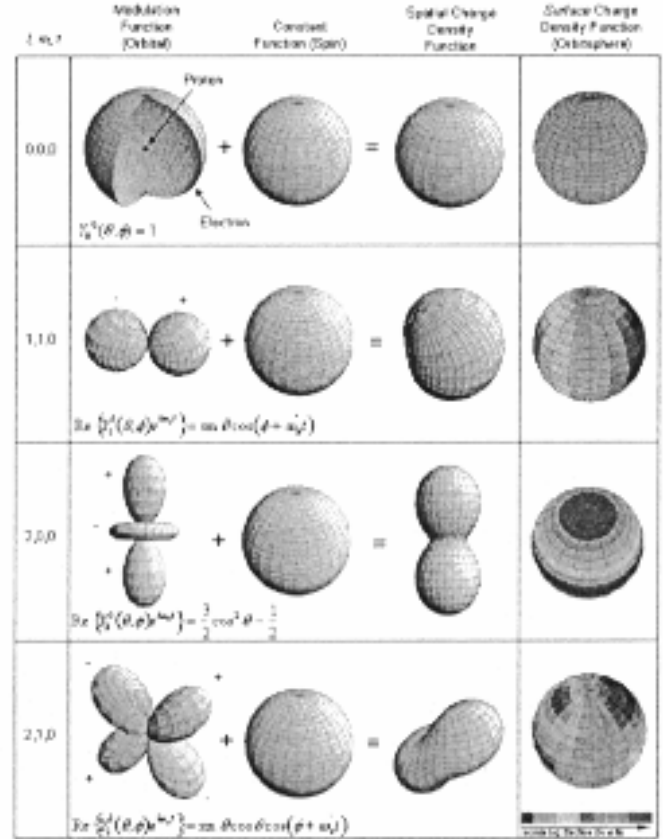


Figure 3. The orbital function modulates the constant (spin) function (shown for $t = 0$; three-dimensional view).

the rotation of two orthogonal great circles about the $(\mathbf{i}_x, \mathbf{i}_y, 0\mathbf{i}_z)$ axis by π wherein one basis-element great circle is initially in the yz plane and the other is in the xz plane:

Step 1:

$$\begin{bmatrix} x' \\ y' \\ z' \end{bmatrix} = \begin{bmatrix} \frac{1}{2} + \frac{\cos \theta}{2} & \frac{1}{2} - \frac{\cos \theta}{2} & -\frac{\sin \theta}{\sqrt{2}} \\ \frac{1}{2} - \frac{\cos \theta}{2} & \frac{1}{2} + \frac{\cos \theta}{2} & \frac{\sin \theta}{\sqrt{2}} \\ \frac{\sin \theta}{\sqrt{2}} & -\frac{\sin \theta}{\sqrt{2}} & \cos \theta \end{bmatrix} \cdot \left(\begin{bmatrix} 0 \\ r_n \cos \phi \\ r_n \sin \phi \end{bmatrix} + \begin{bmatrix} r_n \cos \phi \\ 0 \\ r_n \sin \phi \end{bmatrix} \right). \quad (10)$$

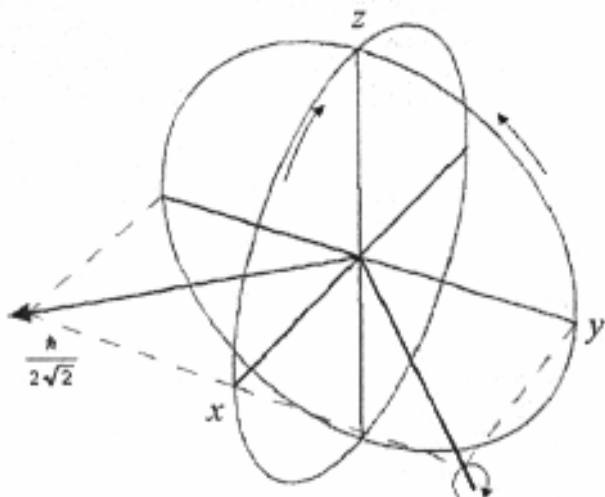


Figure 4. The current on the great circle in the $y'z'$ plane moves counterclockwise and the current on the great circle in the $x'z'$ plane moves clockwise. The xyz system is the laboratory frame, and the orthogonal-current-loop basis set is rigid with respect to the $x'y'z'$ system that rotates about the $(\mathbf{i}_x, \mathbf{i}_y, 0\mathbf{i}_z)$ axis by π radians to generate the elements of the first component of the orbitsphere-cvf. The angular momentum of the orthogonal great circle current loops in the $x'y'$ plane that is evenly distributed over the surface is $\hbar/2\sqrt{2}$.

The first component of the orbitsphere-cvf given by (10) can also be generated by each of rotating a great circle basis element initially in the yz or the xz plane about the $(\mathbf{i}_x, \mathbf{i}_y, 0\mathbf{i}_z)$ axis by 2π radians, as shown in Figs. 5 and 6, respectively.

The orthogonal great circle basis set for step 2 is shown in Fig. 7. The second half of the orbitsphere-cvf, the orbitsphere-cvf component of step 2, is generated by the rotation of two orthogonal great circles about the $(-\mathbf{i}_x/\sqrt{2}, \mathbf{i}_y/\sqrt{2}, \mathbf{i}_z)$ axis by π , wherein one basis-element great circle is initially in the plane that bisects the xy quadrant and is parallel to the z axis and the other is in the xy plane:

Step 2:

$$\begin{bmatrix} x' \\ y' \\ z' \end{bmatrix} = [\mathbf{u} \ \mathbf{v} \ \mathbf{w}] \cdot \left(\begin{bmatrix} \frac{r_n \cos \phi}{\sqrt{2}} \\ r_n \cos \phi \\ \frac{r_n \sin \phi}{\sqrt{2}} \end{bmatrix} + \begin{bmatrix} r_n \cos \phi \\ r_n \sin \phi \\ 0 \end{bmatrix} \right), \quad (11)$$

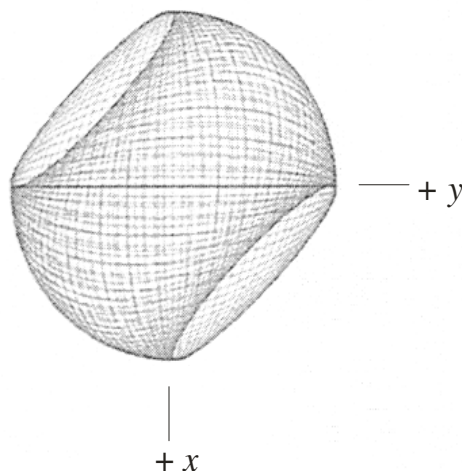


Figure 5. The current pattern of the orbitsphere-cvf component of step 1 shown with 6° increments of θ from the perspective of looking along the z axis. The yz plane great circle current loop that served as a basis element was initially in the yz plane.

where

$$\mathbf{u} = \begin{bmatrix} \frac{1+3\cos\theta}{4} \\ \frac{-1+\cos\theta-2\sqrt{2}\sin\theta}{4} \\ \frac{1}{2} \left(\frac{-1+\cos\theta}{\sqrt{2}} + \sin\theta \right) \end{bmatrix},$$

$$\mathbf{v} = \begin{bmatrix} \frac{-1+\cos\theta+2\sqrt{2}\sin\theta}{4} \\ \frac{1+3\cos\theta}{4} \\ \frac{\sqrt{2}-\sqrt{2}\cos\theta+2\sin\theta}{4} \end{bmatrix},$$

$$\mathbf{w} = \begin{bmatrix} \frac{-\sqrt{2}+\sqrt{2}\cos\theta-2\sin\theta}{4} \\ \frac{\sqrt{2}-\sqrt{2}\cos\theta-2\sin\theta}{4} \\ \cos^2 \frac{\theta}{2} \end{bmatrix}.$$

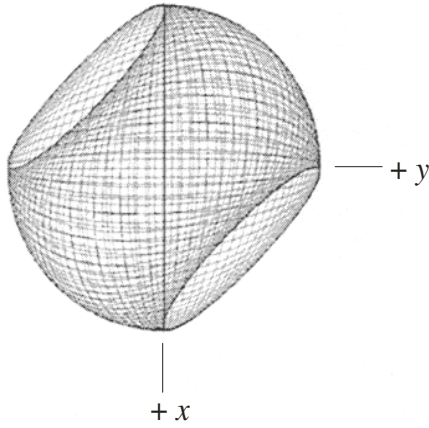


Figure 6. The current pattern of the orbitsphere-cvf component of step 1 shown with 6° increments of θ from the perspective of looking along the z axis. The great circle current loop that served as a basis element was initially in the xz plane.

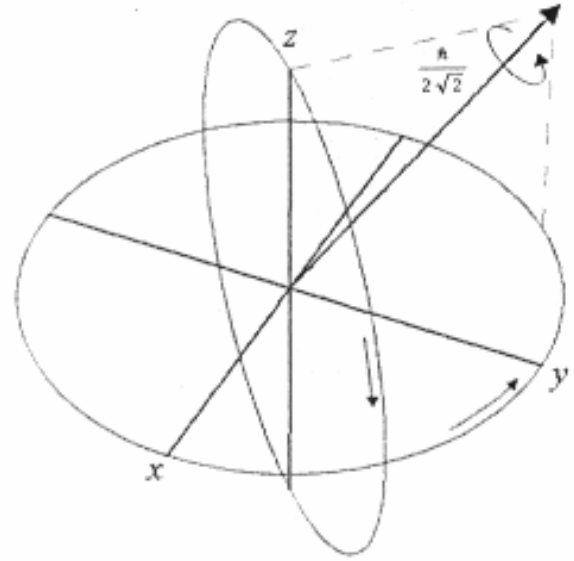


Figure 7. The current on the great circle in the plane that bisects the $x'y'$ quadrant and is parallel to the z' axis moves clockwise, and the current on the great circle in the $x'y'$ plane moves counterclockwise. Rotation of the great circles about the $(-\mathbf{i}_x/\sqrt{2}, \mathbf{i}_y/\sqrt{2}, \mathbf{i}_z)$ axis by π radians generates the elements of the second component of the orbitsphere-cvf. The angular momentum vector along the $(-\mathbf{i}_x/\sqrt{2}, \mathbf{i}_y/\sqrt{2}, \mathbf{i}_z)$ z axis is $\hbar/2\sqrt{2}$, corresponding to each of the z and $-xy$ components of magnitude $\hbar/4$.

The second component of the orbitsphere-cvf given by (11) can also be generated by each of rotating a great circle basis element that is initially in the plane that bisects the xy quadrant and is parallel to the z axis or is in the xy plane about the $(-\mathbf{i}_x/\sqrt{2}, \mathbf{i}_y/\sqrt{2}, \mathbf{i}_z)$ axis by 2π radians, as shown in Figs. 8 and 9, respectively.

The orbitsphere-cvf is given by the superposition of the components from step 1 and from step 2. The current pattern of the orbitsphere-cvf generated by the rotations of the orthogonal great circle current loops is a continuous and total coverage of the spherical surface, but it is shown as visual representations using 6° increments θ in Figs. 2A–C.

The resultant angular momentum projections of $\mathbf{L}_{xy} = \hbar/4$ and $\mathbf{L}_z = \hbar/2$ from the convolution operator meet the boundary condition for the unique current with an angular velocity magnitude at each point on the surface given by (6) and give rise to the Stern–Gerlach experiment, as shown in Ref. 7. Specifically, he further constraint that the current density is uniform such that the charge density is uniform, corresponding to an equipotential, minimum-energy surface, is satisfied by using the orbitsphere-cvf as a basis element to generate $Y_0^0(\phi, \theta)$ using a convolution operator comprising an autocorrelation-type function, as given in Appendix III of Ref. 7. The operator comprises the convolution of each great circle current loop of the orbitsphere-cvf, designated as the primary orbitsphere-cvf, with a second orbitsphere-cvf designated as the secondary orbitsphere-cvf,

wherein the convolved secondary elements are matched for orientation, angular momentum, and phase to those of the primary. The resulting exact uniform current distribution obtained from the convolution has the angular momentum distribution, with components of $\mathbf{L}_{xy} = \hbar/4$ and $\mathbf{L}_z = \hbar/2$.

3.3 Angular Functions

The time, radial, and angular solutions of the wave equation are separable. Also, based on the radial solution, the angular charge- and current-density functions of the electron, $A(\theta, \phi, t)$, must be a solution of the wave equation in two dimensions (plus time),

$$\left[\nabla^2 - \frac{1}{v^2} \frac{\partial^2}{\partial t^2} \right] A(\theta, \phi, t) = 0, \quad (12)$$

where $\rho(r, \theta, \phi, t) = f(r)A(\theta, \phi, t) = 1/r^2 \delta(r - r_n)A(\theta, \phi, t)$ and $A(\theta, \phi, t) = Y(\theta, \phi)k(t)$; thus

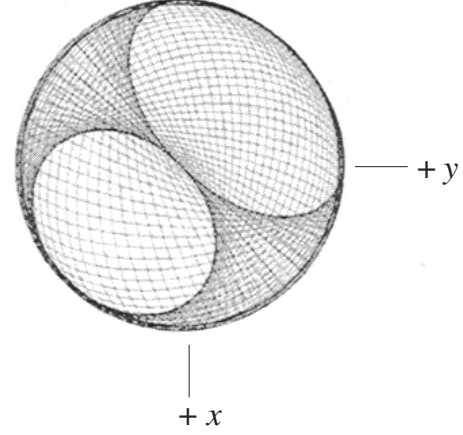
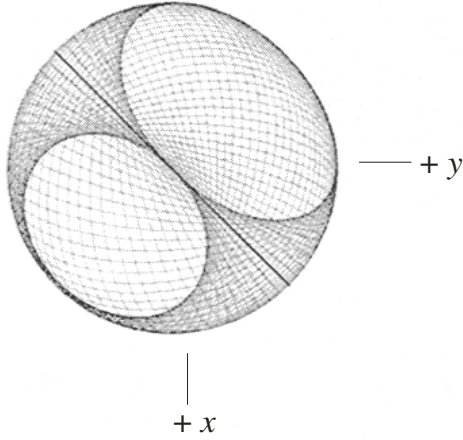


Figure 8. The current pattern of the orbitsphere-cvf component of step 2 shown with 6° increments of θ from the perspective of looking along the z axis. The great circle current loop that served as a basis element was initially in the plane that bisects the xy quadrant and was parallel to the z axis.

$$\left[\frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) \right]_{r,\phi} + \frac{1}{r^2 \sin^2 \theta} \left(\frac{\partial^2}{\partial \phi^2} \right)_{r,\theta} - \frac{1}{v^2} \frac{\partial^2}{\partial t^2} \right] A(\theta, \phi, t) = 0, \quad (13)$$

where v is the linear velocity of the electron. The charge-density functions, including the time-function factor, are, for $\ell = 0$,

$$\rho(r, \theta, \phi, t) = \frac{e}{8\pi r^2} [\delta(r - r_n)] [Y_0^0(\theta, \phi) + Y_\ell^m(\theta, \phi)], \quad (14)$$

and, for $\ell \neq 0$,

$$\rho(r, \theta, \phi, t) = \frac{e}{4\pi r^2} [\delta(r - r_n)] [Y_0^0(\theta, \phi) + \text{Re}\{Y_\ell^m(\theta, \phi)e^{i\omega_n t}\}], \quad (15)$$

where $Y_\ell^m(\theta, \phi)$ are the spherically harmonic functions that spin about the z axis with angular frequency ω_n , with $Y_0^0(\theta, \phi)$ the constant function. $\text{Re}\{Y_\ell^m(\theta, \phi)e^{i\omega_n t}\} = P_\ell^m(\cos \theta)\cos(m\phi + \omega_n t)$, where to keep the form of the spherically harmonic as a traveling wave about the z axis, $\omega_n' = m\omega_n$.

Figure 9. The current pattern of the orbitsphere-cvf component of step 2 shown with 6° increments of θ from the perspective of looking along the z axis. The great circle current loop that served as a basis element was initially in the xy plane.

3.4 Acceleration without Radiation

3.4.1 Special-Relativistic Correction to the Electron Radius

The relationship between the electron wavelength and its radius is given by (4), where λ is the de Broglie wavelength. For each current-density element of the spin function the distance along each great circle in the direction of instantaneous motion undergoes length contraction and time dilation. Using a phase-matching condition, the wavelengths of the electron and laboratory inertial frames are equated, and the corrected radius is given by

$$r_n = r_n' \left[\sqrt{1 - \left(\frac{v}{c}\right)^2} \sin \left[\frac{\pi}{2} \left(1 - \left(\frac{v}{c}\right)^2\right)^{3/2} \right] + \frac{1}{2\pi} \cos \left[\frac{\pi}{2} \left(1 - \left(\frac{v}{c}\right)^2\right)^{3/2} \right] \right], \quad (16)$$

where the electron velocity is given by (6). (See Ref. 7, Chapter 1, Special Relativistic Correction to the Ionization Energies section). e/m_e of the electron, the electron angular momentum of \hbar , and μ_B are invariant, but the mass and charge densities increase in the laboratory frame due to the relativistically contracted electron radius. As $v \rightarrow c$, $r/r' \rightarrow 1/(2\pi)$ and $r = \lambda$, as shown in Fig. 10.

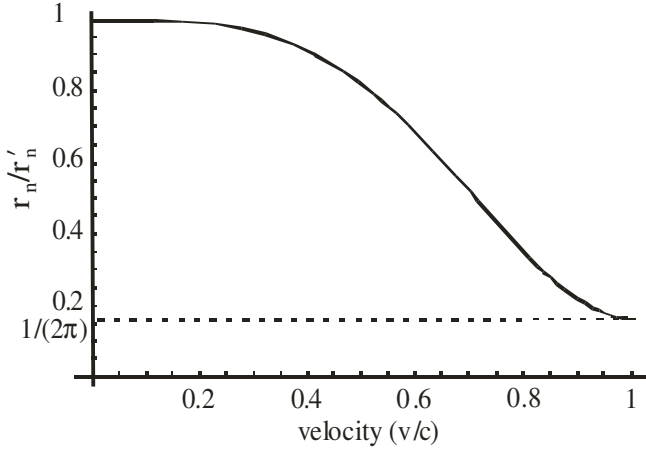


Figure 10. The normalized radius as a function of the velocity due to relativistic contraction (16).

3.4.2 Nonradiation Based on the Space-Time Fourier Transform of the Electron Current

The Fourier transform of the electron charge-density function given by (8) is a solution of the three-dimensional wave equation in frequency space (\mathbf{k} , ω space), as given in Chapter 1, Spacetime Fourier Transform of the Electron Function section, of Ref. 7. Then the corresponding Fourier transform of the current-density function $K(s, \Theta, \Phi, \omega)$ is given by multiplying by the constant angular frequency:

$$\begin{aligned}
 K(s, \Theta, \Phi, \omega) &= 4\pi\omega_n \frac{\sin(2s_n r_n)}{2s_n r_n} \\
 &\otimes 2\pi \sum_{\nu=1}^{\infty} \frac{(-1)^{\nu-1} (\pi \sin \Theta)^{2(\nu-1)}}{(\nu-1)!(\nu-1)!} \\
 &\frac{\Gamma\left(\frac{1}{2}\right)\Gamma\left(\nu+\frac{1}{2}\right)}{(\pi \cos \Theta)^{2\nu+1} 2^{\nu+1}} \frac{2\nu!}{(\nu-1)!} \frac{1}{s^{2\nu}} \\
 &\otimes 2\pi \sum_{\nu=1}^{\infty} \frac{(-1)^{\nu-1} (\pi \sin \Phi)^{2(\nu-1)}}{(\nu-1)!(\nu-1)!} \\
 &\frac{\Gamma\left(\frac{1}{2}\right)\Gamma\left(\nu+\frac{1}{2}\right)}{(\pi \cos \Phi)^{2\nu+1} 2^{\nu+1}} \frac{2\nu!}{(\nu-1)!} \frac{1}{s^{2\nu}} \frac{\delta(\omega-\omega_n)+\delta(\omega+\omega_n)}{4\pi}.
 \end{aligned} \tag{17}$$

$\mathbf{s}_n \cdot \mathbf{v}_n = \mathbf{s}_n \cdot \mathbf{c} = \omega_n$ implies $r_n = \lambda_n$, which is given by (16) in the case that k is the light-like k^0 . In this case, (17) vanishes. Consequently, space-time harmonics of $\omega_n/c = k$ or $\omega_n/c(\mathcal{E}\epsilon_0)^{1/2} = k$ for which the Fourier transform of the current-density function is nonzero

do not exist. Radiation due to charge motion does not occur in any medium when this boundary condition is met. Nonradiation is also determined directly from the fields based on Maxwell's equations, as given in Section 3.4.3.

3.4.3 Nonradiation Based on the Electron Electromagnetic Fields and the Poynting Power Vector

A point charge undergoing periodic motion accelerates and as a consequence radiates according to the Larmor formula

$$P = \frac{1}{4\pi\epsilon_0} \frac{2e^2}{3c^3} a^2, \tag{18}$$

where e is the charge, a is its acceleration, ϵ_0 is the permittivity of free space, and c is the speed of light. Although an accelerated *point* particle radiates, an *extended distribution* modeled as a superposition of accelerating charges does not have to radiate.^(37,39,42-44)

In Ref. 3 and Appendix I, Chapter 1, of Ref. 7, the electromagnetic far field is determined from the current distribution in order to obtain the condition, if it exists, that the electron current distribution must satisfy such that the electron does not radiate. The current follows from (14)–(15). The currents corresponding to (14) and the first term of (15) are static. Thus they are trivially nonradiative. The current due to the time-dependent term of (15) corresponding to p , d , f , etc., orbitals is

$$\begin{aligned}
 \mathbf{J} &= \frac{\omega_n}{2\pi} \frac{e}{4\pi r_n^2} N[\delta(r-r_n)] \text{Re}\{Y_\ell^m(\theta, \phi)\} [\mathbf{u}(t) \times \mathbf{r}] \\
 &= \frac{\omega_n}{2\pi} \frac{e}{4\pi r_n^2} N'[\delta(r-r_n)] \\
 &\times (P_\ell^m(\cos \theta) \cos(m\phi + \omega'_n t)) [\mathbf{u} \times \mathbf{r}] \\
 &= \frac{\omega_n}{2\pi} \frac{e}{4\pi r_n^2} N'[\delta(r-r_n)] \\
 &\times (P_\ell^m(\cos \theta) \cos(m\phi + \omega'_n t)) \sin \theta \hat{\phi},
 \end{aligned} \tag{19}$$

where to keep the form of the spherical harmonic as a traveling wave about the z axis, $\omega'_n = m\omega_n$ and N and N' are normalization constants. The vectors are defined as

$$\hat{\phi} = \frac{\hat{u} \times \hat{r}}{|\hat{u} \times \hat{r}|} = \frac{\hat{u} \times \hat{r}}{\sin \theta}, \quad \hat{u} = \hat{z} = \text{orbital axis}, \tag{20}$$

$$\hat{\boldsymbol{\theta}} = \hat{\boldsymbol{\phi}} \times \hat{\boldsymbol{r}}. \quad (21)$$

“ \wedge ” denotes the unit vectors $\hat{\boldsymbol{u}} \equiv \mathbf{u}/|\mathbf{u}|$, non-unit vectors are in bold, and the current function is normalized. For the electron source current given by (19), each comprising a multipole of order (ℓ, m) with a time dependence $e^{i\omega_n t}$, the far-field solutions to Maxwell’s equations are given by

$$\begin{aligned} \mathbf{B} &= -\frac{i}{k} a_M(\ell, m) \nabla \times g_\ell(kr) \mathbf{X}_{\ell, m}, \\ \mathbf{E} &= a_M(\ell, m) g_\ell(kr) \mathbf{X}_{\ell, m}, \end{aligned} \quad (22)$$

and the time-averaged power radiated per solid angle $dP(\ell, m)/d\Omega$ is

$$\frac{dP(\ell, m)}{d\Omega} = \frac{c}{8\pi k^2} |a_M(\ell, m)|^2 |\mathbf{X}_{\ell, m}|^2, \quad (23)$$

where

$$a_M(\ell, m) = \frac{-ek^2}{c\sqrt{\ell(\ell+1)}} \frac{\omega_n}{2\pi} Nj_\ell(kr_n) \Theta \sin(mks). \quad (24)$$

In the case that k is the light-like k^0 , then $k = \omega_n/c$, in (24), and (22)–(23) vanish for

$$s = vT_n = R = r_n = \lambda_n. \quad (25)$$

There is no radiation.

3.5 Magnetic Field Equations of the Electron

The orbitsphere is a shell of negative charge current comprising correlated charge motion along great circles. For $\ell = 0$ the orbitsphere gives rise to a magnetic moment of one Bohr magneton⁽⁴⁵⁾ (the details of the derivation of the magnetic parameters, including the electron g factor, are given in Ref. 3 and Chapter 1 of Ref. 7):

$$\mu_B = \frac{e\hbar}{2m_e} = 9.274 \times 10^{-24} \text{ JT}^{-1}. \quad (26)$$

The magnetic field of the electron shown in Fig. 11 is given by

$$\mathbf{H} = \frac{e\hbar}{m_e r_n^3} (\mathbf{i}_r \cos \theta - \mathbf{i}_\theta \sin \theta) \text{ for } r < r_n, \quad (27)$$

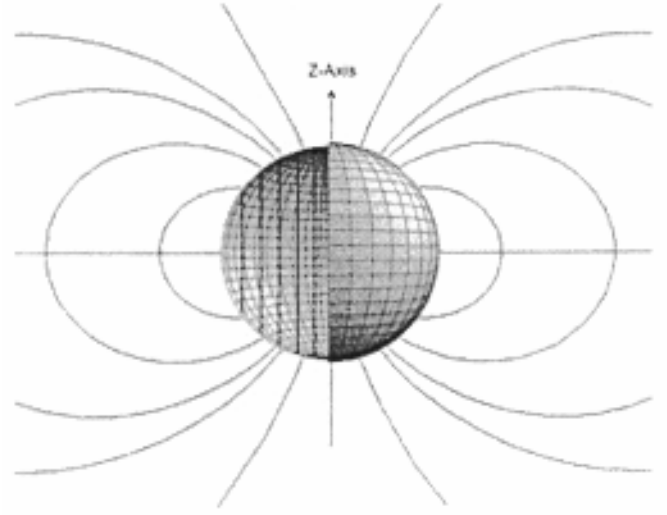


Figure 11. The magnetic field of an electron orbitsphere (z axis defined as the vertical axis).

$$\mathbf{H} = \frac{e\hbar}{2m_e r^3} (\mathbf{i}_r 2 \cos \theta + \mathbf{i}_\theta \sin \theta) \text{ for } r > r_n. \quad (28)$$

The energy stored in the magnetic field of the electron is

$$E_{mag} = \frac{1}{2} \mu_0 \int_0^{2\pi} \int_0^\pi \int_0^\infty H^2 r^2 \sin \theta dr d\theta d\Phi, \quad (29)$$

$$E_{mag \text{ total}} = \frac{\pi \mu_0 e^2 \hbar^2}{m_e^2 r_1^3} = \frac{4\pi \mu_0 \mu_B^2}{r_1^3}. \quad (30)$$

3.6 Stern–Gerlach Experiment

The Stern–Gerlach experiment implies a magnetic moment of one Bohr magneton and an associated angular momentum quantum number of $1/2$. Historically, this quantum number has been called the spin quantum number, s ($s = 1/2$, $m_s = \pm 1/2$). The superposition of the vector projection of the orbitsphere angular momentum on the z axis is $\hbar/2$, with an orthogonal component of $\hbar/4$. Excitation of a resonant Larmor precession gives rise to \hbar on an axis \mathbf{S} that precesses about the z axis, called the spin axis, at the Larmor frequency at an angle of $\theta = \pi/3$ to give a perpendicular projection of

$$\mathbf{S}_\perp = \hbar \sin \frac{\pi}{3} = \pm \sqrt{\frac{3}{4}} \hbar \mathbf{i}_y \quad (31)$$

and a projection onto the axis of the applied magnetic field of

$$\mathbf{S}_\parallel = \hbar \cos \frac{\pi}{3} = \pm \frac{\hbar}{2} \mathbf{i}_z. \quad (32)$$

The superposition of the $\hbar/2$, z axis component of the orbitosphere angular momentum and the $\hbar/2$, z axis component of \mathbf{S} gives \hbar corresponding to the observed electron magnetic moment of a Bohr magneton, μ_B .

3.7 Electron g Factor

As given in the Electron g Factor section of Refs. 7 and 3, conservation of angular momentum of the orbitosphere permits a discrete change of its “kinetic angular momentum” ($\mathbf{r} \times m\mathbf{v}$) by the applied magnetic field of $\hbar/2$, and concomitantly the “potential angular momentum” ($\mathbf{r} \times e\mathbf{A}$) must change by $-\hbar/2$:

$$\Delta \mathbf{L} = \frac{\hbar}{2} - \mathbf{r} \times e\mathbf{A} \quad (33)$$

$$= \left[\frac{\hbar}{2} - \frac{e\phi}{2\pi} \right] \hat{z}. \quad (34)$$

In order that the change of angular momentum, $\Delta \mathbf{L}$, equal zero, ϕ must be $\Phi_0 = h/(2e)$, the magnetic flux quantum. The magnetic moment of the electron is parallel or antiparallel to the applied field only. During the spin-flip transition, power must be conserved. Power flow is governed by the Poynting power theorem:

$$\nabla \cdot (\mathbf{E} \times \mathbf{H}) = -\frac{\partial}{\partial t} \left[\frac{\mu_0}{2} \mathbf{H} \cdot \mathbf{H} \right] - \frac{\partial}{\partial t} \left[\frac{\epsilon_0}{2} \mathbf{E} \cdot \mathbf{E} \right] - \mathbf{J} \cdot \mathbf{E}. \quad (35)$$

Equation (36) gives the total energy of the flip transition, which is the sum of the energy of reorientation of the magnetic moment (first term), the magnetic energy (second term), the electric energy (third term), and the dissipated energy of a fluxon treading the orbitosphere (fourth term), respectively:

$$\Delta E_{mag}^{spin} = 2 \left(1 + \frac{\alpha}{2\pi} + \frac{2\alpha^2}{3} \left(\frac{\alpha}{2\pi} \right) - \frac{4}{3} \left(\frac{\alpha}{2\pi} \right)^2 \right) \mu_B B, \quad (36)$$

$$\Delta E_{mag}^{spin} = g \mu_B B, \quad (37)$$

where the stored magnetic energy corresponding to the $\partial/\partial t[(1/2)\mu_0 \mathbf{H} \cdot \mathbf{H}]$ term increases, the stored electric energy corresponding to the $\partial/\partial t[(1/2)\epsilon_0 \mathbf{E} \cdot \mathbf{E}]$ term increases, and the $\mathbf{J} \cdot \mathbf{E}$ term is dissipative. The spin-flip transition can be considered as involving a magnetic moment of g times that of a Bohr magneton.

The magnetic moment, m , of (36) is twice that from the gyromagnetic ratio, as given by

$$m = \frac{\text{charge} \cdot \text{angular momentum}}{2 \cdot \text{mass}}. \quad (38)$$

The magnetic moment of the electron is the sum of the component corresponding to the kinetic angular momentum, $\hbar/2$, and the component corresponding to the vector potential angular momentum, $\hbar/2$, (33). The spin-flip transition can be considered as involving a magnetic moment of g times that of a Bohr magneton. The g factor is redesignated the fluxon g factor as opposed to the anomalous g factor, and it is given by (36):

$$\frac{g}{2} = 1 + \frac{\alpha}{2\pi} + \frac{2}{3} \alpha^2 \left(\frac{\alpha}{2\pi} \right) - \frac{4}{3} \left(\frac{\alpha}{2\pi} \right)^2. \quad (39)$$

For $\alpha^{-1} = 137.03604(11)$,⁽⁴⁶⁾

$$\frac{g}{2} = 1.001159652120. \quad (40)$$

The experimental value⁽⁴⁷⁾ is

$$\frac{g}{2} = 1.001159652188(4). \quad (41)$$

The calculated and experimental values are within the propagated error of the fine-structure constant. Different values of the fine-structure constant have been recorded from different experimental techniques, and α^{-1} depends on a circular argument between theory and experiment.⁽³²⁾ One measurement of the fine-structure constant based on the electron g factor is $\alpha_{g_e}^{-1} = 137.036006(20)$.⁽³⁵⁾ This value can be contrasted with equally precise measurements employing solid state techniques such as those based on the Josephson effect⁽⁴⁸⁾ ($\alpha_J^{-1} = 137.035963(15)$) or the quantized Hall effect⁽⁴⁹⁾ ($\alpha_H^{-1} = 137.035300(400)$). A method of determining α^{-1} that depends on the

circular methodology between theory and experiment to a lesser extent is to substitute the independently measured fundamental constants μ_0 , e , c , and h into (71). The following values of the fundamental constants are given by Weast:⁽⁴⁶⁾

$$\mu_0 = 4\pi \times 10^{-7} \text{ Hm}^{-1}, \quad (42)$$

$$e = 1.6021892(46) \times 10^{-19} \text{ C}. \quad (43)$$

$$c = 2.99792458(12) \times 10^8 \text{ ms}^{-1}, \quad (44)$$

$$h = 6.626176(36) \times 10^{-34} \text{ JHz}^{-1}. \quad (45)$$

For these constants

$$\alpha^{-1} = 137.03603(82). \quad (46)$$

Substituting the α^{-1} from (46) into (39) gives

$$\frac{g}{2} = 1.001159652137. \quad (47)$$

The experimental value⁽⁴⁷⁾ is

$$\frac{g}{2} = 1.001159652188(4). \quad (48)$$

The *postulated* QED theory of $g/2$ is based on the determination of the terms of a *postulated* power series in α/π , where each *postulated* virtual particle is a source of *postulated* vacuum polarization that gives rise to a *postulated* term. The algorithm involves scores of *postulated* Feynman diagrams corresponding to thousands of matrices with thousands of integrations per matrix requiring decades to reach a consensus on the “appropriate” *postulated* algorithm to remove the intrinsic infinities. The remarkable agreement between (47) and (48) demonstrates that $g/2$ may be derived in closed form from Maxwell’s equations in a simple straightforward manner that yields a result with 11-figure agreement with experiment — the limit of the experimental capability of the measurement of α directly or the fundamental constants to determine α . In Section 2 of Chapter 1 and Appendix II of Ref. 7 the Maxwellian result is contrasted with the QED algorithm of invoking virtual particles, zero-point fluctuations of the vacuum, and negative energy states of the vacuum. Rather than an infinity of radically different QED

models, an essential feature is that *Maxwellian solutions are unique*.

3.8 Spin and Orbital Parameters

The total function that describes the current motion of each electron orbitsphere is composed of two functions. One function, the spin function, is spatially uniform over the orbitsphere, has a quantized angular velocity independent of angle, and gives rise to spin angular momentum. The other function, the modulation function, can be spatially uniform — in which case there is no orbital angular momentum and the magnetic moment of the electron orbitsphere is one Bohr magneton — or not spatially uniform — in which case there is orbital angular momentum. The modulation function also rotates with a quantized angular velocity.

The spin function of the electron corresponds to the nonradiative $n = 1$, $\ell = 0$ state of atomic hydrogen, which is well known as an s state or orbital. (See Fig. 1.) In cases of orbitals of heavier elements and excited states of one-electron atoms and atoms or ions of heavier elements that have the ℓ quantum number not equal to zero and are not constant as given by (14), the constant spin function is modulated by a time- and spherically harmonic function as given by (15) and shown in Fig. 3. The modulation or traveling charge-density wave corresponds to an orbital angular momentum in addition to a spin angular momentum. These states are typically referred to as p, d, f, etc., orbitals. Application of Haus’s⁽³⁹⁾ condition also predicts nonradiation for a constant spin function modulated by a time- and spherically harmonic orbital function. There is acceleration without radiation, as also shown in Section 3.4.3. (Also see Abbott and Griffiths,⁽⁴³⁾ Goedecke,⁽⁴⁴⁾ and Daboul and Jensen.⁽⁴²⁾) However, in the case that such a state arises as an excited state by photon absorption, it is radiative due to a radial dipole term in its current-density function, since it possesses space-time Fourier transform components synchronous with waves traveling at the speed of light.⁽³⁹⁾ (See Instability of Excited States section of Ref. 7.)

3.8.1 Moment of Inertia and Spin and Rotational Energies

The moments of inertia and the rotational energies as a function of the ℓ quantum number for the solutions of the time-dependent electron charge-density functions (14)–(15) given in Section 3.3 are solved using the rigid rotor equation.⁽⁴¹⁾ The details of the derivations of the results as well as the demonstration that (14)–(15) with the results given below are

solutions of the wave equation are given in Chapter 1, Rotational Parameters of the Electron (Angular Momentum, Rotational Energy, Moment of Inertia) section of Ref. 7.

For $\ell = 0$

$$I_z = I_{spin} = \frac{m_e r_n^2}{2}, \quad (49)$$

$$L_z = I \boldsymbol{\omega}_z = \pm \frac{\hbar}{2}, \quad (50)$$

$$\begin{aligned} E_{rotational} &= E_{rotational, spin} = \frac{1}{2} \left[I_{spin} \left(\frac{\hbar}{m_e r_n^2} \right)^2 \right] \\ &= \frac{1}{2} \left[\frac{m_e r_n^2}{2} \left(\frac{\hbar}{m_e r_n^2} \right)^2 \right] = \frac{1}{4} \left[\frac{\hbar^2}{2I_{spin}} \right], \end{aligned} \quad (51)$$

$$T = \frac{\hbar^2}{2m_e r_n^2}. \quad (52)$$

For $\ell \neq 0$

$$I_{orbital} = m_e r_n^2 \left[\frac{\ell(\ell+1)}{\ell^2 + 2\ell + 1} \right]^{\frac{1}{2}} = m_e r_n^2 \sqrt{\frac{\ell}{\ell+1}}, \quad (53)$$

$$\mathbf{L} = I \boldsymbol{\omega}_z = I_{orbital} \boldsymbol{\omega}_z = m_e r_n^2 \left[\frac{\ell(\ell+1)}{\ell^2 + 2\ell + 1} \right]^{\frac{1}{2}} \boldsymbol{\omega}_z \quad (54)$$

$$= m_e r_n^2 \frac{\hbar}{m_e r_n^2} \sqrt{\frac{\ell}{\ell+1}} = \hbar \sqrt{\frac{\ell}{\ell+1}},$$

$$L_{z total} = L_{z spin} + L_{z orbital}, \quad (55)$$

$$\begin{aligned} E_{rotational orbital} &= \frac{\hbar^2}{2I} \left[\frac{\ell(\ell+1)}{\ell^2 + 2\ell + 1} \right] = \frac{\hbar^2}{2I} \left[\frac{\ell}{\ell+1} \right] \\ &= \frac{\hbar^2}{2m_e r_n^2} \left[\frac{\ell}{\ell+1} \right], \end{aligned} \quad (56)$$

$$\langle L_{z orbital} \rangle = 0, \quad (57)$$

$$\langle E_{rotational orbital} \rangle = 0. \quad (58)$$

The orbital rotational energy arises from a spin function (spin angular momentum) modulated by a spherically harmonic angular function (orbital angular momentum). The time-averaged mechanical angular momentum and rotational energy associated with the wave-equation solution comprising a traveling charge-density wave on the orbisphere is zero, as given in (57) and (58), respectively. Thus the principal levels are degenerate except when a magnetic field is applied. In the case of an excited state, the angular momentum of \hbar is carried by the fields of the trapped photon. The amplitudes that couple to external magnetic and electromagnetic fields are given by (54) and (56), respectively. The rotational energy due to spin is given by (51), and the total kinetic energy is given by (52).

3.9 Force Balance Equation

The radius of the nonradiative ($n = 1$) state is solved using the electromagnetic force equations of Maxwell relating the charge- and mass-density functions wherein the angular momentum of the electron is given by \hbar .⁽⁷⁾ The reduced mass arises naturally from an electrodynamic interaction between the electron and the proton of mass m_p :

$$\frac{m_e}{4\pi r_1^2} \frac{v_1^2}{r_1} = \frac{e}{4\pi r_1^2} \frac{Ze}{4\pi \epsilon_0 r_1^2} - \frac{1}{4\pi r_1^2} \frac{\hbar^2}{m_p r_n^3}, \quad (59)$$

$$r_1 = \frac{a_H}{Z}, \quad (60)$$

where a_H is the radius of the hydrogen atom.

3.10 Energy Calculations

From Maxwell's equations the potential energy V , kinetic energy T , and electric energy or binding energy E_{ele} are

$$\begin{aligned} V &= \frac{-Ze^2}{4\pi \epsilon_0 r_1} = \frac{-Z^2 e^2}{4\pi \epsilon_0 a_H} \\ &= -Z^2 \times 4.3675 \times 10^{-18} \text{ J} \\ &= -Z^2 \times 27.2 \text{ eV}, \end{aligned} \quad (61)$$

$$T = \frac{Z^2 e^2}{8\pi \epsilon_0 a_H} = Z^2 \times 13.59 \text{ eV}, \quad (62)$$

$$T = E_{ele} = -\frac{1}{2} \epsilon_0 \int_{-\infty}^{r_1} \mathbf{E}^2 dv, \text{ where } \mathbf{E} = -\frac{Ze}{4\pi\epsilon_0 r^2}, \quad (63)$$

$$\begin{aligned} E_{ele} &= -\frac{Ze^2}{8\pi\epsilon_0 r_1} = -\frac{Z^2 e^2}{8\pi\epsilon_0 a_H} \\ &= -Z^2 \times 2.1786 \times 10^{-18} \text{ J} \\ &= -Z^2 \times 13.598 \text{ eV.} \end{aligned} \quad (64)$$

The calculated Rydberg constant is $10,967,758 \text{ m}^{-1}$; the experimental Rydberg constant is $10,967,758 \text{ m}^{-1}$. For increasing Z the velocity becomes a significant fraction of the speed of light; thus special-relativistic corrections were included in the calculation of the ionization energies of one-electron atoms that are given in Table I.

3.11 Resonant Line Shape and Lamb Shift

The spectroscopic line-width arises from the classical rise-time bandwidth relationship, and the Lamb shift is due to conservation of energy and linear momentum and arises from the radiation reaction force between the electron and the photon. It follows from the Poynting power theorem (35) with spherical radiation that the transition probabilities are given by the ratio of power and the energy of the transition.⁽⁵²⁾ The hydrogen electric dipole transition probability due to the transient radial current from the initial quantum state n_i, ℓ, m_ℓ, m_s and radius r_{n_i} to the final $n_f, \ell \pm 1, m_\ell, m_s$ and radius r_{n_f} derived in Ref. 7 is

$$\frac{1}{\tau} = \frac{\text{power}}{\text{energy}}, \quad (65)$$

$$\begin{aligned} \frac{1}{\tau} &= \frac{1}{m_e c^2} \frac{\eta}{24\pi} \left(\frac{e\hbar}{m_e a_0^2} \right)^2 \frac{1}{(n_f n_i)^2} \\ &= 2.678 \times 10^9 \frac{1}{(n_f n_i)^2} \text{ s}^{-1}. \end{aligned} \quad (66)$$

This rise-time gives rise to Γ , the spectroscopic line-width. The relationship between the rise-time and the bandwidth is given by Siebert.⁽⁵³⁾

$$\tau^2 = 4 \left[\frac{\int_{-\infty}^{\infty} t^2 h^2(t) dt}{\int_{-\infty}^{\infty} h^2(t) dt} - \left(\frac{\int_{-\infty}^{\infty} t h^2(t) dt}{\int_{-\infty}^{\infty} h^2(t) dt} \right)^2 \right], \quad (67)$$

$$\Gamma^2 = 4 \frac{\int_{-\infty}^{\infty} f^2 |H(f)|^2 df}{\int_{-\infty}^{\infty} |H(f)|^2 df}. \quad (68)$$

Applying the Schwartz inequality, the relationship between the rise-time and the bandwidth is³

$$\tau\Gamma \geq \frac{1}{\pi}. \quad (69)$$

From (66) the line-width is proportional to the ratio of the quantum Hall resistance, h/e^2 , and η , the radiation resistance of free space:

$$\eta = \sqrt{\frac{\mu_0}{\epsilon_0}}. \quad (70)$$

And the quantum Hall resistance given in the Quantum Hall Effect section of Ref. 7 was derived using the Poynting power theorem. Also, from (66), the line-width is proportional to the fine-structure constant, α :

$$\alpha = \frac{1}{4\pi} \sqrt{\frac{\mu_0}{\epsilon_0}} \frac{e^2}{\hbar} = \frac{1}{2} \sqrt{\frac{\mu_0}{\epsilon_0}} \frac{e^2}{h} = \frac{\mu_0 e^2 c}{2h}. \quad (71)$$

During a transition, the total energy of the system decays exponentially. Applying (67) and (68) to the case of exponential decay,

$$h(t) = e^{-\alpha t} u(t) = e^{-\frac{2\pi}{T} t} u(t), \quad (72)$$

$$|H(f)| = \frac{1}{\sqrt{\left(\frac{1}{T}\right)^2 + (2\pi f)^2}}, \quad (73)$$

where the rise-time, τ , is the time required for $h(t)$ of (72) to decay to $1/e$ of its initial value and where the bandwidth, Γ , is the half-power bandwidth, the distance between points at which

$$|H(f)| = \frac{|H(0)|}{\sqrt{2}}. \quad (74)$$

Table I: Relativistically Corrected Ionization Energies for Some One-Electron Atoms

| <i>One-e Atom</i> | <i>Z</i> | <i>β (Eqn. (1.267) of Ref. 7)</i> | <i>Theoretical Ionization Energies (eV) ((60), (64), and Eqn. (1.272) of Ref. 7)</i> | <i>Experimental Ionization Energies (eV)^a</i> | <i>Relative Difference between Experimental and Calculated^b</i> |
|-------------------|----------|--|--|--|--|
| H | 1 | 0.00730 | 13.59847 | 13.59844 | -0.000002 |
| He ⁺ | 2 | 0.01459 | 54.41826 | 54.41778 | -0.000009 |
| Li ²⁺ | 3 | 0.02189 | 122.45637 | 122.45429 | -0.000017 |
| Be ³⁺ | 4 | 0.02919 | 217.72427 | 217.71865 | -0.000026 |
| B ⁴⁺ | 5 | 0.03649 | 340.23871 | 340.2258 | -0.000038 |
| C ⁵⁺ | 6 | 0.04378 | 490.01759 | 489.99334 | -0.000049 |
| N ⁶⁺ | 7 | 0.05108 | 667.08834 | 667.046 | -0.000063 |
| O ⁷⁺ | 8 | 0.05838 | 871.47768 | 871.4101 | -0.000078 |
| F ⁸⁺ | 9 | 0.06568 | 1103.220 | 1103.1176 | -0.000093 |
| Ne ⁹⁺ | 10 | 0.07297 | 1362.348 | 1362.1995 | -0.000109 |
| Na ¹⁰⁺ | 11 | 0.08027 | 1648.910 | 1648.702 | -0.000126 |
| Mg ¹¹⁺ | 12 | 0.08757 | 1962.945 | 1962.665 | -0.000143 |
| Al ¹²⁺ | 13 | 0.09486 | 2304.512 | 2304.141 | -0.000161 |
| Si ¹³⁺ | 14 | 0.10216 | 2673.658 | 2673.182 | -0.000178 |
| P ¹⁴⁺ | 15 | 0.10946 | 3070.451 | 3069.842 | -0.000198 |
| S ¹⁵⁺ | 16 | 0.11676 | 3494.949 | 3494.1892 | -0.000217 |
| Cl ¹⁶⁺ | 17 | 0.12405 | 3947.228 | 3946.296 | -0.000236 |
| Ar ¹⁷⁺ | 18 | 0.13135 | 4427.363 | 4426.2296 | -0.000256 |
| K ¹⁸⁺ | 19 | 0.13865 | 4935.419 | 4934.046 | -0.000278 |
| Ca ¹⁹⁺ | 20 | 0.14595 | 5471.494 | 5469.864 | -0.000298 |
| Sc ²⁰⁺ | 21 | 0.15324 | 6035.681 | 6033.712 | -0.000326 |
| Ti ²¹⁺ | 22 | 0.16054 | 6628.064 | 6625.82 | -0.000339 |
| V ²²⁺ | 23 | 0.16784 | 7248.745 | 7246.12 | -0.000362 |
| Cr ²³⁺ | 24 | 0.17514 | 7897.827 | 7894.81 | -0.000382 |
| Mn ²⁴⁺ | 25 | 0.18243 | 8575.426 | 8571.94 | -0.000407 |
| Fe ²⁵⁺ | 26 | 0.18973 | 9281.650 | 9277.69 | -0.000427 |
| Co ²⁶⁺ | 27 | 0.19703 | 10016.63 | 10012.12 | -0.000450 |
| Ni ²⁷⁺ | 28 | 0.20432 | 10780.48 | 10775.4 | -0.000471 |
| Cu ²⁸⁺ | 29 | 0.21162 | 11573.34 | 11567.617 | -0.000495 |

^a From theoretical calculations, interpolation of H isoelectronic and Rydberg series, and experimental data.^(7,50,51)

^b (Experimental-theoretical)/experimental.

From (67),⁽⁵³⁾

$$\tau = T. \quad (75)$$

From (68),⁽⁵³⁾

$$\Gamma = \frac{1}{\pi T}. \quad (76)$$

From (75) and (76) the relationship between the

rise-time and the bandwidth for exponential decay is

$$\tau\Gamma = \frac{1}{\pi}. \quad (77)$$

Photons obey Bose–Einstein statistics. The emitted radiation, the summation of an ensemble of emitted photons each of an exact frequency and energy given by (87), appears as a wave train with effective length c/Γ . Such a finite pulse of radiation is not exactly

monochromatic but has a frequency spectrum covering an interval of the order Γ . The exact shape of the frequency spectrum is given by the square of the Fourier transform of the electric field. Thus the amplitude spectrum is proportional to

$$\mathbf{E}(\omega) \propto \int_0^{\infty} e^{-\alpha t} e^{-i\omega t} dt = \frac{1}{\alpha_i - i\omega}. \quad (78)$$

The coefficient α_i corresponds to the spectroscopic line-width and also to a shift in frequency that arises from the radiation reaction force between the electron and the photon. The energy radiated per unit frequency interval is therefore

$$\frac{dI(\omega)}{d\omega} = I_0 \frac{\Gamma}{2\pi} \frac{1}{(\omega - \omega_0 - \Delta\omega)^2 + (\Gamma/2)^2}, \quad (79)$$

where I_0 is the total energy radiated. The spectral distribution is called a resonant line shape. The width of the distribution at half-maximum intensity is called the half-width or line-breadth and is equal to Γ . Shown in Fig. 12 is such a spectral line. Because of the reactive effects of radiation, the line is shifted in frequency. The small radiative shift of the energy levels of atoms was first observed by Lamb in 1947⁽²⁴⁾ and is called the Lamb shift in his honor.

The Lamb shift of the $^2P_{1/2}$ state of the hydrogen atom with quantum number $\ell = 1$ is calculated by applying conservation of energy and linear momentum to the emitted photon, electron, and atom. The photon emitted by an excited state atom carries away energy, linear momentum, and angular momentum. The initial and final values of the energies and momenta must be conserved between the atom, the electron, and the photon. (Conservation of angular momentum is used to derive the photon's equation in the Equation of the Photon section of Ref. 7). Consider an isolated atom of mass M with an electron of mass m_e in an excited state level at an energy E and moving with velocity \mathbf{V} along the direction in which the photon is to be emitted (the components of motion perpendicular to this direction remain unaffected by the emission and may be ignored). The energy above the "ground" state at rest is

$$\left(E + \frac{1}{2} M \mathbf{V}^2 \right). \quad (80)$$

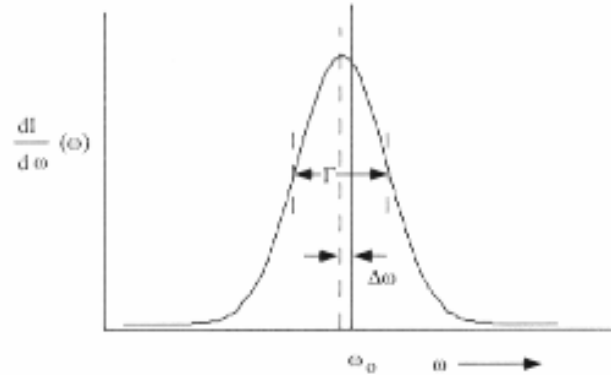


Figure 12. Broadening of the spectral line due to the rise-time and shifting of the spectral line due to the radiative reaction. The resonant line shape has width Γ . The level shift is $\Delta\omega$.

When a photon of energy E_{hv} is emitted, the atom and/or electron recoils and has a new velocity

$$\mathbf{V} + \mathbf{v} \quad (81)$$

(which is a vector sum in that \mathbf{V} and \mathbf{v} may be opposed) and a total energy of

$$\frac{1}{2} M (\mathbf{V} + \mathbf{v})^2. \quad (82)$$

By conservation of energy

$$E + \frac{1}{2} M \mathbf{V}^2 = E_{hv} + \frac{1}{2} M (\mathbf{V} + \mathbf{v})^2, \quad (83)$$

so that the actual energy of the photon emitted is given by

$$E_{hv} = E - \frac{1}{2} M \mathbf{v}^2 - M \mathbf{v} \mathbf{V} = E - E_R - E_D. \quad (84)$$

The photon is thus deficient in energy by a recoil kinetic energy

$$E_R = \frac{1}{2} M \mathbf{v}^2, \quad (85)$$

which is independent of the initial velocity \mathbf{V} , and by a thermal or Doppler energy

$$E_D = M\mathbf{v}\mathbf{V}, \quad (86)$$

which depends on \mathbf{V} ; therefore it can be positive or negative.

Momentum must also be conserved in the emission process. The energy, E , of the photon is given by Planck's equation:

$$E = \hbar\omega = h\frac{\omega}{2\pi} = h\nu = hf = h\frac{c}{\lambda}. \quad (87)$$

From special relativity

$$E = \hbar\omega = mc^2. \quad (88)$$

Thus \mathbf{p} , the momentum of the photon, is

$$\mathbf{p} = mc = \frac{E_{h\nu}}{c}, \quad (89)$$

where c is the velocity of light, so that

$$M\mathbf{V} = M(\mathbf{V} + \mathbf{v}) + \frac{E_{h\nu}}{c}. \quad (90)$$

And the recoil momentum is

$$M\mathbf{v} = -\frac{E_{h\nu}}{c}. \quad (91)$$

Thus the recoil energy is given by

$$E_R = \frac{E_{h\nu}^2}{2Mc^2} \quad (92)$$

and depends on the mass of the electron and/or atom and the energy of the photon. The Doppler energy E_D depends on the thermal motion of the atom and will have a temperature-dependent distribution of values. A mean value, \bar{E}_D , can be defined that is related to the mean kinetic energy per translational degree of freedom^(55,56)

$$\bar{E}_D \cong \frac{1}{2}kT \quad (93)$$

by

$$\bar{E}_D \cong 2\sqrt{E_K E_R} = E_{h\nu} \sqrt{\frac{2\bar{E}_K}{Mc^2}}, \quad (94)$$

where k is Boltzmann's constant and T is the absolute temperature. As a result, the statistical distribution in energy of the emitted photons is displaced from the true excited-state energy by $-E_R$ and broadened by E_D into a Gaussian distribution of width $2\bar{E}_D$. The distribution for absorption has the same shape but is displaced by $+E_R$.

For the transition of the hydrogen atom with $n = 2$ and $\ell = 0$ in the initial and final states, the emitted angular radiation power pattern is uniform. The linear momentum of the photon is balanced by the recoil momentum of the entire atom of mass m_H . The recoil frequency of the hydrogen atom, Δf , is given by combining (87) and (92):

$$\Delta f = \frac{\Delta\omega}{2\pi} = \frac{E_{h\nu}}{h} = \frac{(E_{h\nu})^2}{2hm_H c^2} = 13.3952 \text{ MHz}, \quad (95)$$

where $E_{h\nu}$ corresponding to the recoil energy (92) is

$$\begin{aligned} E_{h\nu} &= 13.5983 \text{ eV} \left(1 - \frac{1}{n^2}\right) - h\Delta f, \\ h\Delta f &\lll 10 \text{ eV}, \quad n = 2; \\ \therefore E_{h\nu} &= 13.5983 \text{ eV} \left(1 - \frac{1}{n^2}\right). \end{aligned} \quad (96)$$

However, during the emission of a photon by an excited-state atom with $\ell \neq 0$, the angular radiation power pattern is not uniform because the charge-density of the electron is not uniform. With $\ell \neq 0$, the charge-density function is a constant function plus a spherically harmonic function (angular modulation) corresponding to spin and orbital angular momenta, respectively, as given in Sections 3.3 and 3.8 and the One-Electron Atom section of Ref. 7. In the case of $\ell = 1$, $m_\ell = 0$ designated the p_x orbital, the angular charge-density function is

$$\begin{aligned} \rho(r, \theta, \phi, t) &= \frac{e}{4\pi r^2} [\delta(r - r_n)] [Y_0^0(\theta, \phi) \\ &\quad + \text{Re}\{Y_\ell^m(\theta, \phi)e^{i\omega_n t}\}], \end{aligned} \quad (97)$$

where $\text{Re}\{Y_\ell^m(\theta, \phi)e^{i\omega_n t}\} = P_\ell^m(\cos \theta)\cos(m\phi + \omega_n t)$ and $\omega_n = 0$ for $m = 0$. Thus

$$Y_{1,z} = \cos \theta. \quad (98)$$

Figure 3 gives a pictorial representation of how the

modulation function changes the electron density on the orbitsphere for several ℓ values. Consequently, rather than the electron recoiling as a point at the origin with transfer of the momentum to the nucleus such that the recoil is with the atom, the electron recoils independently of the atom, and it receives the majority of the recoil momentum according to (92) since the electron mass is 1/1836 times that of the nucleus. The conservation of the momentum between the electron and the photon depends on the angular distribution of the charge density relative to the photon linear propagation axis. Thus the solid angle must be considered. As given in the Equation of the Photon section of Ref. 7, the angular momentum, \mathbf{m} , from the time-averaged angular-momentum density of the emitted photon is given by Eqn. (16.61) of Jackson⁽⁵⁷⁾ in cgs units:

$$\mathbf{m} = \int \frac{1}{8\pi c} \text{Re}[\mathbf{r} \times (\mathbf{E} \times \mathbf{B}^*)] dx^4 = \hbar, \quad (99)$$

where the energy, E , is given from the Poynting power density:⁽⁵⁸⁾

$$E = \int \frac{c}{4\pi} \text{Re}(\mathbf{E} \times \mathbf{H}^*) dx^4 = \hbar \omega. \quad (100)$$

Since the fields have the same multipolarity as the source, from (99)–(100), the radiation power pattern depends on the integral of the angular function squared, $(Y_\ell^m(\theta, \phi))^2$. Specifically, the radiation power pattern of the electron in the ${}^2P_{1/2}$ ($\ell = 1; m_\ell = 0$) state follows from the integral of the square of (98) over the spherical solid angle, as given by McQuarrie:⁽⁴¹⁾

$$\int_0^{2\pi} \int_0^\pi (Y_1^0(\theta, \phi))^2 \sin \theta d\theta d\phi = \frac{4\pi}{3}. \quad (101)$$

The inverse of (101) is the weighting factor of momentum transfer due to the radiation power pattern, as given for antennas after Kong.⁽⁵⁹⁾

The spherical and time harmonics, $\text{Re}\{Y_\ell^m(\theta, \phi)e^{i\omega t}\}$, of the p_x orbital and ${}^2P_{1/2}$ state correspond to a constant current about the z axis. In this case, the photon-momentum transfer for the ${}^2P_{1/2} \rightarrow {}^2S_{1/2}$ transition causes an excitation of a Larmor precession of a vector \mathbf{S} with \hbar of angular momentum about the z axis at an angle of $\theta = \pi/3$, as given in the Spin Angular Momentum of the Orbitsphere-cvf and Orbitsphere ($Y_0^0(\theta, \phi)$ $\ell = 0$ section of Ref. 7. In this case, the orbital angular momentum (Eqn. (2.66) of Ref. 7) is

zero before and after the transition, and the invariance of each of elm_e of the electron, the electron angular momentum of \hbar , and the electron magnetic moment of μ_B from the spin angular momentum is maintained. From Eqn. (1.84) of Ref. 7, the projection of \mathbf{S} onto the transverse plane (xy plane) is $\mathbf{S}_\perp = \hbar \sin(\pi/3) = \pm\sqrt{3}/4 \hbar \mathbf{i}_{y_r}$. Then the photon energy is corrected by the factor $\sqrt{3}/4$ due to electron recoil from the emission of a photon from the ${}^2P_{1/2}$ state corresponding to the rotating transverse component of momentum transfer. In this case, E_{hv} corresponding to the recoil energy (92), including the factor from (101), is

$$E_{hv} = 13.5983 \text{ eV} \left(1 - \frac{1}{n^2}\right) \frac{3}{4\pi} \sqrt{\frac{3}{4}} - \hbar \Delta f, \quad (102)$$

$$\hbar \Delta f \ll 10 \text{ eV}, \quad n = 2;$$

$$\therefore E_{hv} = 13.5983 \text{ eV} \left(1 - \frac{1}{2^2}\right) \frac{3}{4\pi} \sqrt{\frac{3}{4}}.$$

The electron contribution to the Lamb shift of the ${}^2P_{1/2}$ state of the hydrogen atom relative to the higher-energy state ${}^2S_{1/2}$ is given by combining (87), (92), (101), and (102):

$$\Delta f = \frac{\Delta \omega}{2\pi} = \frac{E_{hv}}{h} = \frac{(E_{hv})^2}{2h\mu_e c^2} = 1052.48 \text{ MHz}, \quad (103)$$

wherein the reduced mass of the electron given by Eqn. (1.234) of Ref. 7 corrects for the finite mass of the nucleus during excitation of the precession of \mathbf{S} .⁽⁶⁰⁾ Furthermore, since \mathbf{S} rotates about the z axis at $\theta = \pi/3$, it has a static projection of the angular momentum of $\mathbf{S}_\parallel = \pm \hbar \cos(\pi/3) = \pm (\hbar/2) \mathbf{i}_{z_r}$ as given by Eqn. (1.85) of Ref. 7. The energy and angular momentum of the photon correspond according to (89) and (99)–(100). Therefore the recoil energy of the photon corresponding to momentum transfer to the atom along the z axis for the ${}^2P_{1/2}$ transition is given by the sum of the atom-alone term given by (95) and that due to \mathbf{S}_\parallel , minus the electron recoil term corresponding to \mathbf{S}_\perp :

$$\Delta f = \frac{\Delta \omega}{2\pi} = \frac{E_{hv}}{h} = \frac{(E_{hv})^2}{2hm_H c^2}$$

$$= \frac{\left(13.5983 \text{ eV} \left(1 - \frac{1}{2^2}\right) \left(1 + \frac{1}{2} - \sqrt{\frac{3}{4}}\right)\right)^2}{2hm_H c^2} \quad (104)$$

$$= 5.3839 \text{ MHz}.$$

The momentum of the electron, atom, and photon are conserved. The total recoil energy is the sum of the electron component (103) and the atom component (104). Thus the calculated Lamb shift due to both components of momentum transfer is

$$\begin{aligned}\Delta f &= 1052.48 \text{ MHz} + 5.3839 \text{ MHz} \\ &= 1057.87 \text{ MHz.}\end{aligned}\quad (105)$$

The experimental Lamb shift is $\Delta f = 1057.862 \text{ MHz}$.⁽⁶¹⁾

3.12 Spin-Orbital Coupling (Fine Structure)

The electron's motion in the hydrogen atom is always perpendicular to its radius; consequently, as shown by (7), the electron's angular momentum of \hbar is invariant. Furthermore, the electron is nonradiative due to its angular motion, as shown in Section 3.4. The radiative instability of excited states is due to a radial dipole term in the function representative of the excited state due to the interaction of the photon and the excited-state electron, as shown in the Instability of Excited States section of Ref. 7. The angular momentum of the photon given in the Equation of the Photon section of Ref. 7 is $\mathbf{m} = \int (8\pi c)^{-1} \text{Re}[\mathbf{r} \times (\mathbf{E} \times \mathbf{B}^*)] dx^4 = \hbar$. It is conserved for the solutions of the resonant photons and excited-state electron functions given in the Excited States of the One-Electron Atom (Quantization) section and the Equation of the Photon section of Ref. 7. Thus the electrodynamic angular momentum and the inertial angular momentum are matched such that the correspondence principle holds. It follows from the principle of conservation of angular momentum that e/m_e of the Bohr magneton (26) is invariant (see the Determination of Orbitsphere Radii section of Ref. 7).

A magnetic field is a relativistic effect of the electrical field, as shown by Jackson.⁽⁶²⁾ No energy term is associated with the magnetic field of the electron of the hydrogen atom unless another source of magnetic field is present. In the case of spin-orbit coupling, the invariant \hbar of spin angular momentum and orbital angular momentum each gives rise to a corresponding invariant magnetic moment of a Bohr magneton, and their corresponding energies superimpose, as given in the Orbital and Spin Splitting section of Ref. 7. The interaction of the two magnetic moments gives rise to a relativistic spin-orbit coupling energy. The vector orientations of the momenta must be considered as well as the condition that flux be linked by the electron in units of the magnetic flux quantum in order to conserve the invariant electron angular

momentum of \hbar . The energy may be calculated with the additional conditions of the invariance of the electron's charge and mass to charge ratio e/m_e .

As shown in Section 3.7 and the Electron g Factor section of Ref. 7 (Eqn. (1.160) of Ref. 7 and second right-hand side term of Eqn. (36)), flux must be linked by the electron orbitsphere in units of the magnetic flux quantum that treads the orbitsphere at $v = c$ with a corresponding energy of

$$E_{mag}^{fluxon} = 2 \frac{\alpha}{2\pi} \mu_B B. \quad (106)$$

As shown in the Orbitsphere Equation of Motion for $\ell = 0$ Based on the Current Vector Field (CVF) section of Ref. 7, the maximum projection of the rotating spin angular momentum of the electron onto an axis is $\sqrt{3/4} \hbar$. From (38) the magnetic flux due to the spin angular momentum of the electron is⁽⁴⁵⁾

$$\mathbf{B} = \frac{\mu_0 \mu}{r^3} = \sqrt{\frac{3}{4}} \frac{\mu_0 e \hbar}{2 m_e r^3}, \quad (107)$$

where μ is the magnetic moment. The maximum projection of the orbital angular momentum onto an axis is \hbar , as shown in the Orbital and Spin Splitting section of Ref. 7, with a corresponding magnetic moment of a Bohr magneton μ_B . Substituting the magnetic moment of μ_B corresponding to the orbital angular momentum and (107) for the magnetic flux corresponding to the spin angular momentum into (106) gives the spin-orbit coupling energy $E_{s/o}$:

$$E_{s/o} = 2 \frac{\alpha}{2\pi} \mu_B B = 2 \sqrt{\frac{3}{4}} \frac{\alpha}{2\pi} \left(\frac{e \hbar}{2 m_e} \right) \frac{\mu_0 e \hbar}{2 m_e r^3}. \quad (108)$$

The Bohr magneton corresponding to the orbital angular momentum is invariant and the corresponding invariant electron charge e is common with that which gives rise to the magnetic field due to the spin angular momentum. The condition that the magnetic flux quantum tread the orbitsphere at $v = c$ with the maintenance of the invariance of the electron's mass to charge ratio e/m_e and electron angular momentum of \hbar requires that the radius and the electron mass of the magnetic field term of (108) be relativistically corrected. As shown in Section 3.4.1 and the Space-time Fourier Transform of the Electron Function and

Determination of Orbitosphere Radii sections of Ref. 7, the relativistically corrected radius r^* follows from the relationship between the electron wavelength and the radius:

$$2\pi r = \lambda. \quad (109)$$

As shown in the Excited States of the One-Electron Atom (Quantization) section of Ref. 7, the phase-matching condition requires that the electron wavelength be the same for orbital and spin angular momentum. With $v = c$,

$$r^* = \lambda. \quad (110)$$

Thus

$$r^* = \frac{r}{2\pi}. \quad (111)$$

The relativistically corrected mass m^* follows from (111) with maintenance of the invariance of the electron angular momentum of \hbar given by (6) and (7):

$$m\mathbf{r} \times \mathbf{v} = m_e r \frac{\hbar}{m_e r}. \quad (112)$$

With (111), the relativistically corrected mass m^* is

$$m^* = 2\pi m_e. \quad (113)$$

With the substitution of (111) and (113) into (108), the spin-orbit coupling energy $E_{s/o}$ is given by

$$\begin{aligned} E_{s/o} &= 2 \frac{\alpha}{2\pi} \left(\frac{e\hbar}{2m_e} \right) \frac{\mu_0 e\hbar}{2(2\pi m_e) \left(\frac{r}{2\pi} \right)^3} \sqrt{\frac{3}{4}} \\ &= \sqrt{\frac{3}{4}} \frac{\alpha\pi\mu_0 e^2 \hbar^2}{m_e^2 r^3}. \end{aligned} \quad (114)$$

(The magnetic field in this case is equivalent to that of a point electron at the origin with $\sqrt{3/4} \hbar$ of angular momentum.)

In the case that $n = 2$, the radius given by (4) is $r =$

$2a_0$. The predicted energy difference between the ${}^2P_{3/2}$ and ${}^2P_{1/2}$ levels of the hydrogen atom, $E_{s/o}$, given by (114) is

$$E_{s/o} = \sqrt{\frac{3}{4}} \frac{\alpha\pi\mu_0 e^2 \hbar^2}{8m_e^2 a_0^3}, \quad (115)$$

wherein $\ell = 1$ and both levels are equivalently Lamb shifted.

$E_{s/o}$ may be expressed in terms of the mass-energy of the electron. The energy stored in the magnetic field of the electron orbitosphere (30) is

$$E_{mag} = \frac{\pi\mu_0 e^2 \hbar^2}{(m_e)^2 r_n^3}. \quad (116)$$

As shown in the Pair Production section of Ref. 7 with the $v = c$ condition, the result of the substitution of $\alpha a_0 = \tilde{\lambda}_C$ for r_n and the relativistic mass, $2\pi m_e$, for m_e , and multiplication by the relativistic correction, α^{-1} , which arises from Gauss's law surface integral and the relativistic invariance of charge, is

$$E_{mag} = m_e c^2. \quad (117)$$

Thus (115) can be expressed as

$$\begin{aligned} E_{s/o} &= \frac{\alpha^5 (2\pi)^2}{8} m_e c^2 \sqrt{\frac{3}{4}} \\ &= 4.51905 \times 10^{-5} \text{ eV}. \end{aligned} \quad (118)$$

Using the Planck equation, the corresponding frequency, $\Delta f_{s/o}$, is

$$\Delta f_{s/o} = 10,927.0 \text{ MHz}. \quad (119)$$

As in the case of the ${}^2P_{1/2} \rightarrow {}^2S_{1/2}$ transition, the photon-momentum transfer for the ${}^2P_{3/2} \rightarrow {}^2P_{1/2}$ transition causes an excitation of a Larmor precession of a vector \mathbf{S} with \hbar of angular momentum about the z axis at an angle of $\theta = \pi/3$, as given in the Spin Angular Momentum of the Orbitosphere-cvf and Orbitosphere ($Y_0^0(\theta, \phi)$) $\ell = 0$ section of Ref. 7. In addition, $\Delta m_\ell = -1$; then the photon energy is corrected by the factor $1 - \sqrt{3/4}$ due to electron recoil from the emission of a photon from the ${}^2P_{3/2}$ state corresponding to the rotating transverse component of momentum transfer. In this case, from (102), E_{hv} is

$$E_{hv} = 13.5984 \text{ eV} \left(1 - \frac{1}{2^2}\right) \frac{3}{4\pi} \left(1 - \sqrt{\frac{3}{4}}\right) \quad (120)$$

$$= 0.326196 \text{ eV.}$$

The electron contribution to the recoil shift of the ${}^2P_{3/2} \rightarrow {}^2P_{1/2}$ transition given by (103) and (120) is

$$\Delta f = \frac{\Delta\omega}{2\pi} = \frac{E_{hv}}{h} = \frac{(E_{hv})^2}{2h\mu_e c^2}$$

$$= \frac{\left(13.5984 \text{ eV} \left(1 - \frac{1}{2^2}\right) \frac{3}{4\pi} \left(1 - \sqrt{\frac{3}{4}}\right)\right)^2}{2h\mu_e c^2} \quad (121)$$

$$= 25.1883 \text{ MHz.}$$

Furthermore, $\Delta m_\ell = -1$ corresponds to the static angular momentum change of $\hbar \mathbf{i}_{zR}$. Therefore, from (104), the recoil energy of the photon corresponding to momentum transfer to the atom along the z axis for the ${}^2P_{3/2} \rightarrow {}^2P_{1/2}$ transition due to the atom-alone term given by (95) and that due to $\Delta m_\ell = -1$ minus the electron recoil term corresponding to \mathbf{S}_\perp is

$$\Delta f = \frac{\Delta\omega}{2\pi} = \frac{E_{hv}}{h} = \frac{(E_{hv})^2}{2hm_H c^2}$$

$$= \frac{\left(13.5983 \text{ eV} \left(1 - \frac{1}{2^2}\right) \left(1 + \left(1 - \sqrt{\frac{3}{4}}\right)\right)\right)^2}{2hm_H c^2} \quad (122)$$

$$= 17.2249 \text{ MHz.}$$

The momentum of the electron, atom, and photon are conserved. The total recoil energy is the sum of the electron component (121) and the atom component (122). Thus the calculated recoil frequency, Δf_R , shift due to both components of momentum transfer is

$$\Delta f_R = 25.1883 \text{ MHz} + 17.2249 \text{ MHz} \quad (123)$$

$$= 42.4132 \text{ MHz,}$$

corresponding to an energy, E_R , of

$$E_R = 1.75407 \times 10^{-7} \text{ eV.} \quad (124)$$

The energy, E_{FS} , and frequency, Δf_{FS} , for the ${}^2P_{3/2} \rightarrow {}^2P_{1/2}$ transition called the fine-structure splitting is

given by the sum of (118), (121), and (122) and the sum of (119) and (123), respectively:

$$E_{FS} = \frac{\alpha^5 (2\pi)^2}{8} m_e c^2 \sqrt{\frac{3}{4}}$$

$$+ \left(13.5983 \text{ eV} \left(1 - \frac{1}{2^2}\right)\right)^2$$

$$\times \left[\frac{\left(\frac{3}{4\pi} \left(1 - \sqrt{\frac{3}{4}}\right)\right)^2}{2h\mu_e c^2} + \frac{\left(1 + \left(1 - \sqrt{\frac{3}{4}}\right)\right)^2}{2hm_H c^2} \right] \quad (125)$$

$$= 4.5190 \times 10^{-5} \text{ eV} + 1.75407 \times 10^{-7} \text{ eV}$$

$$= 4.53659 \times 10^{-5} \text{ eV,}$$

$$\Delta f_{FS} = 10,927.0 \text{ MHz} + 42.4132 \text{ MHz} \quad (126)$$

$$= 10,969.4 \text{ MHz.}$$

The energy of $4.53659 \times 10^{-5} \text{ eV}$ corresponds to a frequency of 10,969.4 MHz given by (125) and (126), respectively, or a wavelength of 2.73298 cm. The experimental value of the ${}^2P_{3/2} \rightarrow {}^2P_{1/2}$ transition frequency is 10,969.1 MHz.^(61,63) The large natural widths of the hydrogen 2p levels limits the experimental accuracy,⁽⁶³⁾ yet, given this limitation, the agreement between the theoretical and experimental fine structure is excellent and within the cited errors.

3.13 Spin-Nuclear Coupling (Hyperfine Structure)

The radius of the hydrogen atom is increased or decreased very slightly due to the Lorentzian force on the electron due to the magnetic field of the proton and its orientation relative to the electron's angular momentum vector. The additional small centripetal magnetic force is the relativistic corrected Lorentzian force, \mathbf{F}_{mag} , as also given in the Two-Electron Atom and Three, Four, Five, Six, Seven, Eight, Nine, Ten, Eleven through and Twenty-Electron Atoms sections of Ref. 7.

The orbitsphere with $\ell = 0$ is a shell of negative charge current comprising correlated charge motion along great circles. The superposition of the vector projection of the orbitsphere angular momentum on the z axis is $\mathbf{L}_z = \hbar/2$ (Eqn. (1.77) of Ref. 7), with an orthogonal component of $\mathbf{L}_{xy} = \hbar/4$ (Eqn. (1.76) of Ref. 7). The magnetic field of the electron at the nucleus due to \mathbf{L}_z , after McQuarrie,⁽⁴⁵⁾ is

$$\mathbf{B} = \frac{\mu_0 e \hbar}{2m_e r^3}, \quad (127)$$

where μ_0 is the permeability of free space ($4\pi \times 10^{-7}$ N/A²). An electrodynamic force or radiation reaction force, a force dependent on the second derivative of the charge's position with respect to time, arises between the electron and the proton. This force, given in Sections 6.6, 12.10, and 17.3 of Jackson,⁽⁶⁴⁾ achieves the condition that the sum of the mechanical momentum and electromagnetic momentum is conserved.

The magnetic moment of the proton, μ_p , aligns in the direction of \mathbf{L}_z , but experiences a torque due to the orthogonal component \mathbf{L}_{xy} . As shown in the Orbitsphere Equation of Motion for $\ell = 0$ Based on the Current Vector Field (CVF) section of Ref. 7, the magnetic field of the orbitsphere gives rise to the precession of the magnetic moment vector of the proton directed from the origin of the orbitsphere at an angle of $\theta = \pi/3$ relative to the z axis. The precession of μ_p forms a cone in the nonrotating laboratory frame to give a perpendicular projection of

$$\mu_{p\perp} = \pm \sqrt{\frac{3}{4}} \mu_p, \quad (128)$$

after Eqn. (1.84) of Ref. 7, and a projection onto the z axis of

$$\mu_{p\parallel} = \pm \frac{\mu_p}{2}, \quad (129)$$

after Eqn. (1.85) of Ref. 7. At torque balance \mathbf{L}_{xy} also precesses about the z axis at 90° with respect to $\mu_{p\parallel}$. Using (127), the magnitude of the force F_{mag} between the antiparallel field of the electron and μ_p is

$$F_{mag} = \left| \frac{\mu_p \times \mathbf{B}}{r} \right| = \mu_p \frac{\mu_0 e \hbar}{2m_e r^4}. \quad (130)$$

The radiation reaction force corresponding to photon emission or absorption is radial, as given in the Equation of the Electric Field inside the Orbitsphere section of Ref. 7. The reaction force on the electron due to the force of the electron's field on the magnetic moment of the proton is the corresponding relativistic central force, \mathbf{F}_{mag} , which acts uniformly on each charge- (mass-) density element of the electron. The magnetic central force is derived as follows from the

Lorentzian force, which is relativistically corrected. The Lorentzian force at each point of the electron moving at velocity \mathbf{v} due to a magnetic flux \mathbf{B} is

$$\mathbf{F}_{mag} = e\mathbf{v} \times \mathbf{B}. \quad (131)$$

Equations (130) and (131) may be expressed in terms of the electron velocity given by (6):

$$F_{mag} = \frac{\hbar}{m_e r} \frac{e\mu_0\mu_p}{2r^3} = \frac{e}{2} |\mathbf{v} \times \mathbf{B}|, \quad (132)$$

where \mathbf{B} is the magnetic flux of the proton at the electron. (The magnetic moment \mathbf{m} of the proton is $\mathbf{m} = \mu_p \mathbf{i}_z$, and the magnetic field of the proton follows from the relationship between the magnetic dipole field and the magnetic moment \mathbf{m} , as given by Jackson.⁽⁶⁵⁾) In the light-like frame the velocity \mathbf{v} is the speed of light, and \mathbf{B} corresponds to the time-dependent component of the proton magnetic moment given by (128). Thus the central force is

$$\mathbf{F}_{mag} = \pm \frac{e\alpha c}{2} \frac{\mu_0}{r^3} \mu_p \sqrt{\frac{3}{4}}, \quad (133)$$

where the relativistic factor from Eqn. (1.228) of Ref. 7 is α , the plus corresponds to antiparallel alignment of the magnetic moments of the electron and proton, and the minus corresponds to parallel alignment. From (59) the outward centrifugal force (Eqn. (1.220) of Ref. 7) on the electron is balanced by the electric force (Eqn. (1.221) of Ref. 7) and the magnetic forces given by Eqn. (1.231) of Ref. 7 and (133):

$$\frac{m_e v^2}{r} = \frac{e^2}{4\pi\epsilon_0 r^2} - \frac{\hbar^2}{mr^3} \pm \sqrt{\frac{3}{4}} \frac{e\alpha c}{2} \frac{\mu_0}{r^3} \mu_p. \quad (134)$$

Using (6),

$$\frac{\hbar^2}{m_e r^3} = \frac{e^2}{4\pi\epsilon_0 r^2} - \frac{\hbar^2}{mr^3} \pm \sqrt{\frac{3}{4}} \frac{e\alpha c}{2} \frac{\mu_0}{r^3} \mu_p, \quad (135)$$

$$\frac{\hbar^2}{m_e r^3} + \frac{\hbar^2}{mr^3} = \frac{e^2}{4\pi\epsilon_0 r^2} \pm \sqrt{\frac{3}{4}} \frac{e\alpha c}{2} \frac{\mu_0}{r^3} \mu_p, \quad (136)$$

$$\frac{\hbar^2}{\mu_e} \pm \sqrt{\frac{3}{4}} \frac{e\alpha c \mu_0 \mu_p}{2} = \frac{e^2}{4\pi\epsilon_0} r, \quad (137)$$

$$r = a_H \pm \sqrt{\frac{3}{4} \frac{4\pi\epsilon_0}{2e^2}} e\alpha c \mu_0 \mu_P, \quad (138)$$

$$r = a_H \pm \sqrt{\frac{3}{4} \frac{2\pi\alpha\mu_P}{ec}}, \quad (139)$$

where μ_e is the electron reduced mass given by Eqn. (1.234) of Ref. 7, a_H is the radius of the hydrogen atom given by Eqn. (1.238) of Ref. 7, the plus corresponds to parallel alignment of the magnetic moments of the electron and proton, and the minus corresponds to antiparallel alignment.

3.13.1 Energy Calculations

The magnetic energy to flip the orientation of the proton's magnetic moment, μ_P , from antiparallel to parallel to the direction of the magnetic flux \mathbf{B}_s of the electron (180° rotation of the magnet moment vector), given by the first term of (36), and (127), and (128), is

$$\begin{aligned} \Delta E_{mag}^{proton\ spin} &= -\frac{\mu_0 e \hbar}{2m_e} \mu_P \sqrt{\frac{3}{4} \left(\frac{1}{r_+^3} + \frac{1}{r_-^3} \right)} \\ &= -\mu_0 \mu_B \mu_P \sqrt{\frac{3}{4} \left(\frac{1}{r_+^3} + \frac{1}{r_-^3} \right)} = -1.918365 \times 10^{-24} \text{ J}, \end{aligned} \quad (140)$$

where the Bohr magneton, μ_B , is given by (26).

The change in the electric energy of the electron due to the slight shift in its radius is given by the difference between the electric energies associated with the two possible orientations of the magnetic moment of the electron with respect to the magnetic moment of the proton, parallel versus antiparallel. Each electric energy is given by substituting the corresponding radius given by (139) into (64). The change in electric energy for the flip from antiparallel to parallel alignment, $\Delta E_{ele}^{S/N}$, is

$$\Delta E_{ele}^{S/N} = \frac{-e^2}{8\pi\epsilon_0} \left[\frac{1}{r_+} - \frac{1}{r_-} \right] = 9.597048 \times 10^{-25} \text{ J}. \quad (141)$$

In addition, the interaction of the magnetic moments of the electron and proton increases the magnetic energy, E_{mag} , of the electron given by (30). The term of E_{mag} for the hyperfine structure of the hydrogen atom is similar to that of muonium, given by (161) in Section 3.14:

$$\begin{aligned} E_{mag} &= - \left(1 + \left(\frac{2}{3} \right)^2 + \alpha \left(\cos \frac{\pi}{3} \right)^2 \right) \\ &\quad \times \frac{\pi \mu_0 e^2 \hbar^2}{m_e^2} \left(\frac{1}{r_+^3} - \frac{1}{r_-^3} \right) \\ &= - \left(1 + \left(\frac{2}{3} \right)^2 + \frac{\alpha}{4} \right) 4\pi \mu_0 \mu_B^2 \left(\frac{1}{r_+^3} - \frac{1}{r_-^3} \right) \\ &= 1.748861 \times 10^{-26} \text{ J}, \end{aligned} \quad (142)$$

where the contribution corresponding to electron spin gives the first term, 1, and the second term, $(2/3)^2$, corresponds to the rotation of the electron about the z axis corresponding to the precession of \mathbf{L}_{xy} . The geometrical factor of $2/3$ for the rotation is given in the Derivation of the Magnetic Field section in Chapter 1 (Eqn. (1.119)) of Ref. 7, and the energy is proportional to the magnetic field strength squared, according to (29). The relativistic factor from Eqns. (1.228) and (1.140) and (2.166) of Ref. 7 is α times $(\cos(\pi/3))^2$, where the latter term is due to the nuclear magnetic moment oriented $\theta = \pi/3$ relative to the z axis. The energy is proportional to the magnetic field strength squared, according to (29).

The total energy of the transition from antiparallel to parallel alignment, $\Delta E_{total}^{S/N}$, is given as the sum of (140)–(142):

$$\begin{aligned} \Delta E_{total}^{S/N} &= \Delta E_{mag}^{proton\ spin} + \Delta E_{ele}^{S/N} + E_{mag} \\ &= -1.918365 \times 10^{-24} \text{ J} + 9.597048 \times 10^{-25} \text{ J} \\ &\quad + 1.748861 \times 10^{-26} \text{ J} = -9.411714 \times 10^{-25} \text{ J} \end{aligned} \quad (143)$$

The energy is expressed in terms of wavelength using the Planck relationship, (87):

$$\lambda = \frac{hc}{\Delta E_{total}^{S/N}} = 21.10610 \text{ cm}. \quad (144)$$

The experimental value from the hydrogen maser is 21.10611 cm.⁽⁶⁶⁾ The 21 cm line is important in astronomy for determining the presence of hydrogen. The remarkable agreement between the calculated and experimental values of the hyperfine structure is only limited by the accuracy of the fundamental constants in (139)–(142).

3.14 Muonium Hyperfine Structure Interval

Muonium ($\mu^+ e^-$, M) is the hydrogen-like bound state of a positive muon and an electron. The solution

of the ground state ($1^2S_{1/2}$) hyperfine structure interval of muonium, $\Delta\nu_{Mu}$, is similar to that of the hydrogen atom. The electron binds to the muon as both form concentric orbitspheres with a minimization of energy. From (59) and (134) the outward centrifugal force (Eqn. (1.220) of Ref. 7) on the electron is balanced by the electric force (Eqn. (1.221) of Ref. 7) and the magnetic forces due to the inner positive muon given by Eqn. (1.231) of Ref. 7 and (133). The resulting force balance equation is the same as that for the hydrogen atom given by (134) with the muon mass, m_μ , replacing the proton mass, m , and the muon magnetic moment, μ_μ , replacing the proton magnetic moment, μ_P . The radius of the electron, r_2 , is given by

$$\frac{m_e v^2}{r_2} = \frac{e^2}{4\pi\epsilon_0 r_2^2} - \frac{\hbar^2}{m_\mu r_2^3} \pm \sqrt{\frac{3}{4}} \frac{e\alpha c}{2} \frac{\mu_0}{r_2^3} \mu_\mu. \quad (145)$$

Using (6),

$$\frac{\hbar^2}{m_e r_2^3} = \frac{e^2}{4\pi\epsilon_0 r_2^2} - \frac{\hbar^2}{m_\mu r_2^3} \pm \sqrt{\frac{3}{4}} \frac{e\alpha c}{2} \frac{\mu_0}{r_2^3} \mu_\mu, \quad (146)$$

$$\frac{\hbar^2}{\mu_{e,\mu}} \pm \sqrt{\frac{3}{4}} \frac{e\alpha c \mu_0 \mu_\mu}{2} = \frac{e^2}{4\pi\epsilon_0} r_2, \quad (147)$$

$$r_2 = a_\mu \pm \sqrt{\frac{3}{4}} \frac{2\pi\alpha\mu_\mu}{ec}, \quad (148)$$

$$\begin{aligned} r_{2+} &= a_\mu + \sqrt{\frac{3}{4}} \frac{2\pi\alpha\mu_\mu}{ec} \\ &= 5.31736859 \times 10^{-11} \text{ m}, \end{aligned} \quad (149)$$

$$\begin{aligned} r_{2-} &= a_\mu - \sqrt{\frac{3}{4}} \frac{2\pi\alpha\mu_\mu}{ec} \\ &= 5.31736116 \times 10^{-11} \text{ m}, \end{aligned} \quad (150)$$

where $\mu_{e,\mu}$ is the muonium electron reduced mass given by Eqn. (1.234) of Ref. 7, with the mass of the proton replaced by the mass of the muon, and a_μ is the Bohr radius of the muonium atom given by (170), with the electron reduced mass, μ_e (Eqn. (1.234) of Ref. 7), replaced by $\mu_{e,\mu}$. The plus sign corresponds to parallel alignment of the magnetic moments of the electron and muon, and the minus sign corresponds to antiparallel alignment.

The radii of the muon, r_1 , in different spin states can be determined from r_2 , the radius of the electron (149)–(150), and the opposing forces on the muon due to the bound electron. The outward centrifugal force (Eqn. (1.220) of Ref. 7) on the muon is balanced by the reaction forces given by (145):

$$\frac{m_\mu v^2}{r_1} = \frac{\hbar^2}{m_e r_2^3} \pm \sqrt{\frac{3}{4}} \frac{e\alpha c}{2} \frac{\mu_0}{r_2^3} \mu_\mu. \quad (151)$$

Using (6),

$$\frac{\hbar^2}{m_\mu r_1^3} = \frac{\hbar^2}{m_e r_2^3} \pm \sqrt{\frac{3}{4}} \frac{e\alpha c}{2} \frac{\mu_0}{r_2^3} \mu_\mu, \quad (152)$$

$$r_2^3 = \left(\frac{m_\mu}{m_e} \pm \sqrt{\frac{3}{4}} \frac{m_\mu e\alpha c}{2\hbar^2} \mu_0 \mu_\mu \right) r_1^3, \quad (153)$$

$$r_1 = \frac{r_2}{\left(\frac{m_\mu}{m_e} \pm \sqrt{\frac{3}{4}} \frac{m_\mu e\alpha c}{2\hbar^2} \mu_0 \mu_\mu \right)^{1/3}}. \quad (154)$$

Using (149)–(150) for r_2 ,

$$r_1 = \frac{a_\mu \pm \sqrt{\frac{3}{4}} \frac{2\pi\alpha\mu_\mu}{ec}}{\left(\frac{m_\mu}{m_e} \pm \sqrt{\frac{3}{4}} \frac{m_\mu e\alpha c}{2\hbar^2} \mu_0 \mu_\mu \right)^{1/3}}, \quad (155)$$

$$r_{1+} = \frac{a_\mu + \sqrt{\frac{3}{4}} \frac{2\pi\alpha\mu_\mu}{ec}}{\left(\frac{m_\mu}{m_e} + \sqrt{\frac{3}{4}} \frac{m_\mu e\alpha c}{2\hbar^2} \mu_0 \mu_\mu \right)^{1/3}} \quad (156)$$

$$= 8.9922565 \times 10^{-12} \text{ m},$$

$$r_{1-} = \frac{a_\mu - \sqrt{\frac{3}{4}} \frac{2\pi\alpha\mu_\mu}{ec}}{\left(\frac{m_\mu}{m_e} - \sqrt{\frac{3}{4}} \frac{m_\mu e\alpha c}{2\hbar^2} \mu_0 \mu_\mu \right)^{1/3}} \quad (157)$$

$$= 8.99224822 \times 10^{-12} \text{ m},$$

where the plus corresponds to parallel alignment of the magnetic moments of the electron and muon, and the minus corresponds to antiparallel alignment.

3.14.1 Energy Calculations

The magnetic energy $\Delta E_{mag}^{spin}(\Delta v_{Mu})$ to flip the orientation of the muon's magnetic moment, μ_μ , from antiparallel to parallel to the direction of the magnetic flux \mathbf{B}_s of the electron (180° rotation of the magnet moment vector) given by (140) is

$$\begin{aligned}\Delta E_{mag}^{spin}(\Delta v_{Mu}) &= -\sqrt{\frac{3}{4}} \frac{\mu_0 e \hbar}{2m_e} \mu_\mu \left(\frac{1}{r_{2+}^3} + \frac{1}{r_{2-}^3} \right) \\ &= -\sqrt{\frac{3}{4}} \mu_0 \mu_B \mu_\mu \left(\frac{1}{r_{2+}^3} + \frac{1}{r_{2-}^3} \right) \quad (158) \\ &= -6.02890320 \times 10^{-24} \text{ J},\end{aligned}$$

wherein the muon magnetic moment replaces the proton magnetic moment, and the electron Bohr magneton, μ_B , is given by (26).

An electric field equivalent to that of a point charge of magnitude $+e$ at the origin only exists for $r_1 < r \leq r_2$. Thus the change in the electric energy of the electron due to the slight shift in its radius is given by the difference between the electric energies associated with the two possible orientations of the magnetic moment of the electron with respect to the magnetic moment of the muon, parallel versus antiparallel. Each electric energy is given by substituting the corresponding radius given by (148) into (64) or (141). The change in electric energy for the flip from antiparallel to parallel alignment is

$$\begin{aligned}\Delta E_{ele}(\Delta v_{Mu}) &= \frac{-e^2}{8\pi\epsilon_0} \left[\frac{1}{r_{2+}} - \frac{1}{r_{2-}} \right] \quad (159) \\ &= 3.02903048 \times 10^{-24} \text{ J}.\end{aligned}$$

For each lepton, applying a magnetic field with a resonant Larmor excitation gives rise to a precessing angular momentum vector \mathbf{S} of magnitude \hbar directed from the origin of the orbitsphere at an angle of $\theta = \pi/3$ relative to the applied magnetic field. As given in the Spin Angular Momentum of the Orbitsphere-cv and Orbitsphere ($Y_0^0(\theta, \phi)$) $\ell = 0$ section of Ref. 7, \mathbf{S} rotates about the axis of the applied field at the Larmor frequency. The magnitude of the components of \mathbf{S} that are parallel and orthogonal to the applied field (Eqns. (1.84)–(1.85) of Ref. 7) are $\hbar/2$ and $\sqrt{3}/4 \hbar$, respectively. Since both the radio frequency

(RF) field and the orthogonal components shown in Fig. 1.15 of Ref. 7 rotate at the Larmor frequency, the RF field that causes a Stern–Gerlach transition produces a stationary magnetic field with respect to these components, as described by Patz.⁽⁶⁷⁾ The corresponding central field at the orbitsphere surface given by the superposition of the central field of the lepton and that of the photon follows from Eqns. (2.10)–(2.17) and Eqn. (17) of Box 1.3 of Ref. 7:

$$\begin{aligned}\mathbf{E} &= \frac{e}{4\pi\epsilon_0 r^2} [Y_0^0(\theta, \phi) \mathbf{i}_r \\ &\quad + \text{Re}\{Y_\ell^m(\theta, \phi) e^{i\omega t}\} \mathbf{i}_y \delta(r - r_1)],\end{aligned} \quad (160)$$

where the spherically harmonic dipole $Y_\ell^m(\theta, \phi) = \sin \theta$ is with respect to the \mathbf{S} axis. The dipole spins about the \mathbf{S} axis at the angular velocity given by (9). The resulting current is nonradiative, as shown in Section 3.4. Thus the field in the RF rotating frame is magnetostatic, as shown in Fig. 1.9 of Ref. 7, but directed along the \mathbf{S} axis.

The interaction of the magnetic moments of the leptons increases their magnetic energies, given by (30), with the mass of the corresponding lepton:

$$\begin{aligned}E_{mag,e}^{stored}(\Delta v_{Mu}) &= -\left(1 + \left(\frac{2}{3} \cos \frac{\pi}{3} \right)^2 + \alpha \right) \frac{\pi \mu_0 e^2 \hbar^2}{m_e^2} \left(\frac{1}{r_{2+}^3} - \frac{1}{r_{2-}^3} \right) \quad (161) \\ &= -\left(1 + \left(\frac{2}{3} \cos \frac{\pi}{3} \right)^2 + \alpha \right) 4\pi \mu_0 \mu_B^2 \left(\frac{1}{r_{2+}^3} - \frac{1}{r_{2-}^3} \right) \\ &= 4.23209178 \times 10^{-26} \text{ J},\end{aligned}$$

$$\begin{aligned}E_{mag,\mu}^{stored}(\Delta v_{Mu}) &= -\left(1 + \left(\frac{2}{3} \cos \frac{\pi}{3} \right)^2 + \alpha \right) \frac{\pi \mu_0 e^2 \hbar^2}{m_\mu^2} \left(\frac{1}{r_{1+}^3} - \frac{1}{r_{1-}^3} \right) \quad (162) \\ &= -\left(1 + \left(\frac{2}{3} \cos \frac{\pi}{3} \right)^2 + \alpha \right) 4\pi \mu_0 \mu_{B,\mu}^2 \left(\frac{1}{r_{1+}^3} - \frac{1}{r_{1-}^3} \right) \\ &= 1.36122030 \times 10^{-28} \text{ J},\end{aligned}$$

where (i) the radii of the electron and muon are given by (149)–(150) and (156)–(157), respectively, (ii) $\mu_{B,\mu}$ is the muon Bohr magneton given by (26) with the electron mass replaced by the muon mass, (iii) the

first term is due to lepton spin, (iv) the second term, $[(2/3)\cos(\pi/3)]^2$, is due to \mathbf{S} , oriented $\theta = \pi/3$ relative to the z axis, wherein the geometrical factor of $2/3$ corresponds to the source current of the dipole field (Eqn. (160)) given in the Derivation of the Magnetic Field section in Chapter 1 (Eqn. (1.119)) of Ref. 7, and the energy is proportional to the magnetic field strength squared, according to (29), and (v) the relativistic factor from Eqns. (1.228) and (1.140) and (2.166) of Ref. 7 is α .

The energy of the ground state ($1^2S_{1/2}$) hyperfine structure interval of muonium, $\Delta E(\Delta\nu_{Mu})$, is given by the sum of (158)–(159) and (161)–(162):

$$\begin{aligned}\Delta E(\Delta\nu_{Mu}) &= \Delta E_{mag}^{spin}(\Delta\nu_{Mu}) + \Delta E_{ele}(\Delta\nu_{Mu}) \\ &+ E_{mag,e}^{stored}(\Delta\nu_{Mu}) + E_{mag,\mu}^{stored}(\Delta\nu_{Mu}) \\ &= -6.02890320 \times 10^{-24} \text{ J} + 3.02903048 \times 10^{-24} \text{ J} \quad (163) \\ &+ 4.23209178 \times 10^{-26} \text{ J} + 1.36122030 \times 10^{-28} \text{ J} \\ &= -2.95741568 \times 10^{-24} \text{ J}.\end{aligned}$$

Using Planck's equation (87), the interval frequency, $\Delta\nu_{Mu}$, and wavelength, $\Delta\lambda_{Mu}$, are

$$\Delta\nu_{Mu} = 4.46330328 \text{ GHz}, \quad (164)$$

$$\Delta\lambda_{Mu} = 6.71682919 \text{ cm}. \quad (165)$$

The experimental hyperfine structure interval of muonium⁽⁶⁸⁾ is

$$\begin{aligned}\Delta E(\Delta\nu_{Mu}) &= -2.957415336 \times 10^{-24} \text{ J}, \\ \Delta\nu_{Mu} &= 4.463302765(53) \text{ GHz} \quad (12 \text{ ppm}), \quad (166) \\ \Delta\lambda_{Mu} &= 6.71682998 \text{ cm}.\end{aligned}$$

There is remarkable (seven to eight significant figure) agreement between the calculated and experimental values of $\Delta\nu_{Mu}$ that is only limited by the accuracy of the fundamental constants in (156)–(159) and (161)–(162), as shown by using different CODATA values.^(46,69,70)

3.15 Positronium

Pair production, the creation of a positron/electron pair, occurs such that the radius of one orbitsphere is infinitesimally greater than the radius of the antiparticle orbitsphere, as discussed in the Pair Production section and the Leptons section of Ref. 7. In addition, a minimum energy may be obtained by the binding of

a positron and an electron as concentric orbitspheres at the same radius such that the electric fields mutually cancel with the conservation of \hbar of angular momentum of each lepton. The short-lived hydrogen-like atom comprising an electron and a positron is called positronium. Before annihilation, positronium can exist with the electron and positron spins parallel or antiparallel, called orthopositronium (3S_1) and parapositronium (1S_0), respectively. Due to the opposite charge of the positron, the magnetic moments are opposed to the spin orientations. The respective decay times are 1 ns and 1 μ s. The splitting of the spectral lines due to spin orientations is called the hyperfine structure of positronium.

The forces of positronium are central, and the radius of the outer orbitsphere (electron or positron) is calculated as follows after (59), (134), and (145). The centrifugal force is given by Eqn. (1.220) of Ref. 7. The centripetal electric force of the inner orbitsphere on the outer orbitsphere is given by Eqn. (1.221) of Ref. 7. A second centripetal force is the relativistic corrected magnetic force, \mathbf{F}_{mag} , between each point of the particle and the antiparticle given by Eqn. (1.231) of Ref. 7, with m_e substituted for m . The force balance equation is given by Eqn. (1.232) of Ref. 7, with m_e substituted for m . The balance between the centrifugal and electric and magnetic forces is given in the Excited States of the One-Electron Atom (Quantization) section and the Excited States of Helium section of Ref. 7 and Ref. 6:

$$\frac{m_e v^2}{r_1} = \frac{\hbar^2}{m_e r_1^3} = \frac{e^2}{4\pi\epsilon_0 r_1^2} - \frac{\hbar^2}{m_e r_1^3}, \quad (167)$$

$$r_1 = \frac{4\pi\epsilon_0 \hbar^2}{e^2 \mu}, \quad (168)$$

where $r_1 = r_2$ is the radius of the positron and the electron and where the reduced mass, μ , is

$$\mu = \frac{m_e}{2}. \quad (169)$$

The Bohr radius given by

$$a_0 = \frac{4\pi\epsilon_0 \hbar^2}{e^2 m_e} \quad (170)$$

and (169) is substituted into (168) to give the ground-state radius of positronium:

$$r_1 = 2a_0. \quad (171)$$

3.15.1 Excited-State Energies

The potential energy V between the particle and the antiparticle with radius r_1 given by (61) is

$$\begin{aligned} V &= \frac{-e^2}{4\pi\epsilon_0 r_1} = \frac{-Z^2 e^2}{8\pi\epsilon_0 a_0} = -2.18375 \times 10^{-18} \text{ J} \\ &= 13.59 \text{ eV}. \end{aligned} \quad (172)$$

The calculated ionization energy is $V/2$ (62)–(64), which is

$$E_{ele} = 6.795 \text{ eV}. \quad (173)$$

The experimental ionization energy is 6.795 eV.

Parapositronium, a singlet-state hydrogen-like atom comprising an electron and a positron, can absorb a photon that excites the atom to the first triplet state, orthopositronium. In parapositronium the electron and positron angular momentum vectors are antiparallel, whereas the magnetic moment vectors are parallel. The opposite relationships exist for orthopositronium. The balance between the centrifugal and electric and magnetic forces is

$$\frac{m_e v^2}{r_n} = \frac{\hbar^2}{m_e r_n^3} = \frac{1}{n} \frac{e^2}{4\pi\epsilon_0 r_n^2} - \frac{\hbar^2}{m_e r_n^3}, \quad (174)$$

$$r_n = n2a_0, \quad (175)$$

where n is an integer and both electrons are at the same excited-state radius of $r_n = n2a_0$. The principal energy levels for the singlet excited states are given by Eqns. (2.22) and (9.12) of Ref. 7, with the electron reduced mass (169) substituted for the mass of the electron:

$$E_n = \frac{1}{n} \frac{e^2 \mu}{8\pi\epsilon_0 r_n} = \frac{1}{n} \frac{e^2 \mu}{8\pi\epsilon_0 n a_0} = \frac{6.795}{n^2} \text{ eV}. \quad (176)$$

The levels given by (176) match the experimental energy levels.

3.15.2 Hyperfine Structure

As shown in the Spin Angular Momentum of the Orbitsphere-cvf and Orbitsphere ($Y_0^0(\theta, \phi)$) $\ell = 0$ section of Ref. 7, the angular momentum of the electron or positron orbitsphere in a magnetic field

comprises the initial $\hbar/2$ projection on the z axis and the initial $\hbar/4$ vector component in the xy plane that precesses about the z axis. As further shown in the Magnetic Parameters of the Electron (Bohr Magneton) section of Ref. 7, a resonant excitation of the Larmor precession frequency gives rise to an additional component of angular momentum that is consistent with Maxwell's equations. As shown in the Excited States of the One-Electron Atom (Quantization) section of Ref. 7, conservation of the \hbar of angular momentum of a trapped photon can give rise to \hbar of electron angular momentum along the \mathbf{S} axis. The photon standing waves of excited states are spherically harmonic functions that satisfy Laplace's equation in spherical coordinates and provide the force balance for the corresponding charge- (mass-) density waves. Consider the photon in the case of the precessing electron with a Bohr magneton of magnetic moment along the \mathbf{S} axis. The radius of the orbitsphere is unchanged, and the photon gives rise to current on the surface that satisfies the condition

$$\nabla \cdot \mathbf{J} = 0 \quad (177)$$

corresponding to a rotating spherically harmonic dipole⁽⁷⁾ that phase-matches the current- (mass-) density of Eqn. (1.123) of Ref. 7. Thus the electrostatic energy is constant, and only the magnetic energy need be considered, as given by (179)–(180). The corresponding central field at the orbitsphere surface given by the superposition of the central field of the lepton and that of the photon follows from Eqns. (2.10)–(2.17) and (17) of Box 1.3 of Ref. 7 and is the spherically harmonic dipole with respect to the \mathbf{S} axis given by (160). The dipole spins about the \mathbf{S} axis at the angular velocity given by (9). The resulting current is nonradiative, as shown in Section 3.4. Thus the field in the RF rotating frame is magnetostatic, as shown in Fig. 1.17 of Ref. 7, but directed along the \mathbf{S} axis.

Applying a magnetic field with a resonant Larmor excitation gives rise to a precessing angular momentum vector \mathbf{S} of magnitude \hbar directed from the origin of the orbitsphere at an angle of $\theta = \pi/3$ relative to the applied magnetic field. \mathbf{S} rotates about the axis of the applied field at the Larmor frequency. The magnitude of the components of \mathbf{S} that are parallel and orthogonal to the applied field (Eqns. (1.84)–(1.85) of Ref. 7) are $\hbar/2$ and $\sqrt{3}/4 \hbar$, respectively. Since both the RF field and the orthogonal components shown in Fig. 1.15 of Ref. 7 rotate at the Larmor frequency, the RF field that causes a Stern–Gerlach transition produces a

stationary magnetic field with respect to these components, as described by Patz.⁽⁶⁷⁾

The component of Eqn. (1.85) of Ref. 7 adds to the initial $\hbar/2$ parallel component to give a total of \hbar in the stationary frame corresponding to a Bohr magneton, μ_B , of magnetic moment. The potential energy of a magnetic moment \mathbf{m} in the presence of flux \mathbf{B} ⁽⁴⁵⁾ is

$$E = \mathbf{m} \cdot \mathbf{B}. \quad (178)$$

The angular momentum of the electron gives rise to a magnetic moment of μ_B . Thus the energy ΔE_{mag}^{spin} to switch from parallel to antiparallel to the field is given by Eqn. (1.147) of Ref. 7:

$$\Delta E_{mag}^{spin} = 2\mu_B \mathbf{i}_z \cdot \mathbf{B} = 2\mu_B B \cos \theta = 2\mu_B B. \quad (179)$$

ΔE_{mag}^{spin} is also given by Planck's equation. It can be shown from conservation of angular momentum considerations (Eqns. (26)–(34) of Box 1.3 of Ref. 7) that the Zeeman splitting is given by Planck's equation and the Larmor frequency based on the gyro-magnetic ratio (Eqn. (2) of Box 1.3 of Ref. 7). The electron's magnetic moment may only be parallel or antiparallel to the magnetic field rather than at a continuum of angles including perpendicular, according to (178). No continuum of energies predicted by (178) for a pure magnetic dipole is possible. The energy difference for the magnetic moment to flip from parallel to antiparallel to the applied field is

$$\Delta E_{mag}^{spin} = 2\hbar\omega_L, \quad (180)$$

corresponding to magnetic dipole radiation, wherein ω_L is the Larmor angular frequency.

Equation (178) implies a continuum of energies, whereas Eqn. (29) of Box 1.3 of Ref. 7 shows that the static-kinetic and dynamic vector potential components of the angular momentum are quantized at $\hbar/2$. Consequently, as shown in Section 3.7 and the Electron g Factor section of Ref. 7, the flux linked during a spin transition is quantized as the magnetic flux quantum, $\Phi_0 = h/(2e)$. Only the states corresponding to $m_s = \pm 1/2$ are possible due to conservation of angular momentum. It is further shown using the Poynting power vector, with the requirement that flux be linked in units of the magnetic flux quantum, that the factor 2 of (179) and (180) is replaced by the electron g factor. The energy, ΔE_{mag}^{spin} , to flip the electron's magnetic moment from parallel to antiparallel to the applied field is given by (36)–(37).

Positronium undergoes a Stern–Gerlach transition. The energy of the transition from orthopositronium (3S_1) to parapositronium (1S_0) is the hyperfine structure interval. The angular momentum of the photon given by $\mathbf{m} = [(8\pi c)^{-1} \text{Re}[\mathbf{r} \times (\mathbf{E} \times \mathbf{B}^*)] dx^4 = \hbar$ in the Equations of the Photon section of Ref. 7 is conserved⁽⁵⁷⁾ for the solutions for the resonant photons and hyperfine-state lepton functions, as shown for the cases of one-electron atoms and helium in the Excited States of the One-Electron Atom (Quantization) and Excited States of Helium sections of Ref. 7, respectively, and Ref. 6. To conserve the \hbar of angular momentum of each lepton and the photon, orthopositronium possesses orbital angular momentum states corresponding to $m_\ell = 0, \pm 1$, whereas parapositronium possesses orbital angular momentum states corresponding to the quantum number $m_\ell = 0$. The orbital angular momentum states of orthopositronium are degenerate in the absence of an applied magnetic field. As in the case of the electron Stern–Gerlach transition, both leptons remain at the same radius of $r = 2a_0$ given by (171).

The hyperfine structure interval of positronium can be calculated from the spin-spin and spin-orbital coupling energies of the $^3S_1 \rightarrow ^1S_0$ transition. The vector projection of the orbitsphere angular momentum on the z axis is $\mathbf{L}_z = \hbar/2$ (Eqn. (1.77) of Ref. 7), with an orthogonal component of $\mathbf{L}_{xy} = \hbar/4$ (Eqn. (1.76) of Ref. 7). The magnetic flux, \mathbf{B} , of the electron (positron) at the positron (electron) due to \mathbf{L}_z , after McQuarrie,⁽⁴⁵⁾ is given by (127). The spin-spin coupling energy $\Delta E_{spin-spin}$ between the inner orbitsphere and the outer orbitsphere is given by (37), where μ_B , the magnetic moment of the outer orbitsphere, is given by (26). Substitution of (26) and (127) into (37) gives

$$\begin{aligned} \Delta E_{spin-spin} &= \frac{1}{2} \frac{g\mu_0 e^2 \hbar^2}{4m_e^2 r_1^3} = \frac{g\mu_0 e^2 \hbar^2}{8m_e^2 (2a_0)^3} \\ &= \frac{1}{8\pi\alpha} \frac{g\alpha^5 (2\pi)^2}{8} m_e c^2, \end{aligned} \quad (181)$$

where the factor of 1/2 arises from (178) with the presence of the magnetic flux only for the 1S_0 state, the radius is given by (171), and (115)–(118) were used to convert (181) to the electron mass-energy form of (118).

In the case of atomic hydrogen with $n = 2$, the radius given by Eqn. (2.2) of Ref. 7 is $r = 2a_0$, and the predicted energy difference between the $^2P_{3/2}$ and $^2P_{1/2}$

levels of the hydrogen atom, $E_{s/o}$, is given by (118). In the case of the hyperfine transition of positronium, the spin-orbital coupling energy $\Delta E_{s/o}$ (${}^3S_1 \rightarrow {}^1S_0$) with $r = 2a_0$ is given by (118), with the requirement that the flux from the partner lepton be linked in units of the magnetic flux quantum corresponding to the anomalous g factor (36)–(37), the source current given by (160) give rise to a factor of $3/2$, and each lepton contribute to the energy:

$$\Delta E_{s/o} ({}^3S_1 \rightarrow {}^1S_0) = 2 \frac{3}{2} \frac{g\alpha^5 (2\pi)^2}{8} m_e c^2 \sqrt{\frac{3}{4}}. \quad (182)$$

The hyperfine structure interval of positronium (${}^3S_1 \rightarrow {}^1S_0$) is given by the sum of (181) and (182):

$$\begin{aligned} \Delta E_{Ps \text{ hyperfine}} &= \Delta E_{spin-spin} + \Delta E_{s/o} ({}^3S_1 \rightarrow {}^1S_0) \\ &= \frac{g\mu_0 e^2 \hbar^2}{8m_e^2 (2a_0)^3} + \frac{3g\alpha^5 (2\pi)^2}{8} m_e c^2 \sqrt{\frac{3}{4}} \\ &= \frac{g\alpha^5 (2\pi)^2}{8} m_e c^2 \left(\frac{1}{8\pi\alpha} + \frac{3\sqrt{3}}{2} \right) \\ &= 8.41155110 \times 10^{-4} \text{ eV}. \end{aligned} \quad (183)$$

Using Planck's equation (87), the interval in frequency, $\Delta\nu$, is

$$\Delta\nu = 203.39041 \text{ GHz}. \quad (184)$$

The experimental ground-state hyperfine structure interval⁽⁷¹⁾ is

$$\begin{aligned} \Delta E_{Ps \text{ hyperfine}} (\text{exper.}) &= 8.41143 \times 10^{-4} \text{ eV}, \\ \Delta\nu (\text{exper.}) &= 203.38910(74) \text{ GHz (3.6 ppm)}. \end{aligned} \quad (185)$$

There is remarkable (six significant figure) agreement between the calculated and experimental values of $\Delta\nu$ that is only limited by the accuracy of the fundamental constants.⁽⁷⁰⁾ A computer simulation of positronium and the positron-electron-annihilation event is shown in the Positronium section of Ref. 7.

4. CONCLUSION

It is true that the Schrödinger equation can be solved exactly for the hydrogen atom, although it is not true that the result is the exact solution of the hydrogen atom. Electron spin is missed entirely, and there are many internal inconsistencies and nonphysical

consequences that do not agree with experimental results.^(1–10) Despite its successes, QM has remained mysterious to all who have encountered it. Starting with Bohr and progressing into the present, the departure from intuitive, physical reality has widened. The connection between QM and reality is more than just a “philosophical” issue. It reveals that QM is not a correct or complete theory of the physical world and that inescapable internal inconsistencies and incongruities arise when attempts are made to treat it as a physical as opposed to a purely mathematical “tool.” But QM has severe limitations, even as a tool, considering that beyond one-electron atoms, multi-electron atom quantum-mechanical equations cannot be solved except by approximation methods⁽¹²⁾ involving adjustable-parameter theories that often involve new physics or constructs or are simply curve-fitting algorithms.⁽⁶⁾

Even the Schrödinger equation results for one-electron atoms (the only problem that can be solved without approximations) are not accurate at all. It is nonrelativistic and there are major differences between predicted and experimental ionization energies as Z increases. Furthermore, in addition to spin, it misses the Lamb shift, anomalous magnetic moment of the electron, the fine structure, the hyperfine structure, and spectra of positronium and muonium; it is not stable to radiation; and it has many other problems with predictions that do not match experimentation.^(2–10) It also has an infinite number of solutions, not just the ones given in textbooks, as given in Margenau and Murphy⁽¹¹⁾ and Ref. 9.

The Dirac equation is touted as remedying the non-relativistic nature of the Schrödinger equation and providing an argument for the existence of virtual particles and corresponding so-called QED computer algorithms for calculating unexpected observables such as the Lamb shift and the anomalous magnetic moment of the electron. But both the Schrödinger and Dirac equations have many problems, which make them untenable as representing reality — infinities, lack of Einstein causality (spooky action at a distance), self-interaction, instability to radiation, negative kinetic energy states, Klein paradox, and more.^(1–10) This was argued by the founders of QM.^(22,30,31) Furthermore, QED is completely postulated. It involves a point electron that cannot occupy any volume; consequently, all calculations have “intrinsic infinities” and require renormalization, which is completely arbitrary. It further relies on a string of nonphysical constructs. For example, it is based on postulated polarization of the vacuum by

postulated virtual particles that have no basis in reality, are fantastical at best, and are conclusively shown to be impossible based on special relativity and astrophysical observations.⁽²¹⁾

Rather than invoking renormalization, untestable virtual particles, and polarization of the vacuum by the virtual particles, the results of QED, such as the anomalous magnetic moment of the electron, the Lamb shift, the fine structure and hyperfine structure of the hydrogen atom, and the hyperfine structure intervals of positronium and muonium (thought to be only solvable using QED), are solved exactly from Maxwell's equations.

$(g - 2)/2$ is solved in closed form based on conservation of the electron's angular momentum and the subsequent requirement that flux be linked by the extended electron in quantized units of the magnetic flux quantum $\Phi_0 = h/(2e)$. The Lamb shift is calculated from the conservation of momentum of the emitted photon and the recoiling electron and hydrogen atom. The fine-structure energy is the Lamb-shifted relativistic interaction energy between the spin and orbital magnetic moments due to the corresponding angular momenta. The hyperfine structure of the hydrogen atom and muonium is calculated from the force balance contribution between the electron and the proton and muon, respectively. The transition

energies correspond to the Stern–Gerlach and stored electric and magnetic energy changes. With positronium, the leptons are at the same radius, and the positronium hyperfine interval is given by the sum of the Stern–Gerlach and fine-structure energies. In each case, the agreement is to the limit possible based on experimental measurements and the error of the measured fundamental constants in the closed-form equations containing only these constants. These results from the known physical laws based on direct observation invalidate virtual particles and confirm QED's illegitimacy as representative of reality.

The laws of electromagnetism and electrodynamics summarized in Maxwell's equations predate QM by over 100 years. These laws and the implicit special relativity are the most experimentally proven physical laws ever. Even before the present work, they were known to hold over at least 24 orders of magnitude of length scale.⁽⁷²⁾ It is evident that only theories consistent with Maxwell's equations and special relativity need be considered. In discovering the means to extend these laws to problems thought only solvable using the mechanics of QED, it is shown that Maxwell's equations are fact and the virtual particle-based QED is fiction.

Received 18 May 2005.

Résumé

La déclaration que l'électrodynamique quantique (EDQ) est la théorie la plus réussie de l'histoire subit un examen critique. L'équation de Dirac a été postulée en 1926 en tant que moyen de résoudre la nature non relativiste de l'équation de Schrödinger afin d'offrir le quatrième chiffre quantique manquant. Les termes de racine carrée positive ainsi que négative fournissent un argument pour l'existence des états négatifs d'énergie du vide, de particules virtuelles et des soi-disant algorithmes correspondant EDQ par ordinateur afin de calculer les visibles inattendus tel que le déplacement de Lamb et le moment magnétique anormale de l'électron. En plus, la tentative de Dirac de résoudre l'électron lie physiquement avec la stabilité en respect de la radiation selon les équations de Maxwell avec les contraintes supplémentaires que c'était un invariant relativiste duquel survient le spin électronique peut être réalisée utilisant une approche classique. Commencant avec les mêmes essentiels physiques que Bohr, Schrödinger, et Dirac du eG en mouvement dans le champ Coulombien du proton et l'équation d'onde, les avancées de la compréhension de la stabilité de l'électron lié à la radiation est appliquée afin de résoudre pour la nature précise de un électron. Plutôt qu'utiliser la condition au limites postulée de Schrödinger, « $\Psi \rightarrow 0$ as $r \rightarrow 64$ », menant à un modèle purement mathématique de l'électron, la contrainte est basée sur l'observation expérimentale. Utilisant les équations de Maxwell, l'équation classique d'onde est résolue avec la contrainte que l'électron en état lié, $n - 1$, ne peut pas émettre d'énergie. En dépit du fait qu'il soit bien connu

qu'une particule ponctuelle accélérée radie, une distribution étendue modelée en tant qu'une superposition de charges en accélération peut ne pas radier. Un modèle invariant physique simple surgit naturellement dans lequel les résultats prévus sont extrêmement directs et intérieurement cohérents exigeant une mathématique minimale comme dans le cas des équations les mieux connues de Newton, Maxwell, Einstein, de Broglie, et Planck sur lesquelles les modèles sont basés. Aucune nouvelle physique est nécessaire, seulement les lois physiques connues basées sur l'observation directe sont utilisées. Plutôt que d'invoquer des théories imaginaires qui ne peuvent pas être soumises à l'essai, les résultats de EDQ tels que le déplacement de Lamb et le moment magnétique anormale de électron, la structure fine de l'atome d'hydrogène et les intervalles de la structure hyperfine du positonium et du muonium peuvent être résolus exactement des équations de Maxwell à la limite possible base sur les mesures expérimentales ce qui confirme l'illégitimité du EDQ en tant qu'expression de la réalité.

Endnotes

- ¹ The Rutherford experiment demonstrated that even atoms comprise essentially empty space.⁽²⁰⁾ Zero-point field fluctuations, virtual particles, and states of negative energy and mass invoked to describe the vacuum are nonsensical and have no basis in reality since they have never been observed experimentally and would correspond to an essentially infinite cosmological constant throughout the entire universe, including regions of no mass. As given by Waldrop,⁽²¹⁾ “What makes this problem into something more than metaphysics is that the cosmological constant is observationally zero to a very high degree of accuracy. And yet, ordinary quantum field theory predicts that it ought to be enormous, about 120 orders of magnitude larger than the best observational limit. Moreover, this prediction is almost inescapable because it is a straightforward application of the uncertainty principle, which in this case states that every quantum field contains a certain, irreducible amount of energy even in empty space. Electrons, photons, quarks — the quantum field of every particle contributes. And that energy is exactly equivalent to the kind of pressure described by the cosmological constant. The cosmological constant has accordingly been an embarrassment and a frustration to every physicist who has ever grappled with it.”
- ² Oskar Klein pointed out a glaring paradox implied by the Dirac equation, which was never resolved:⁽²³⁾ “Electrons may penetrate an electrostatic barrier even when their kinetic energy, $E - mc^2$, is lower than the barrier. Since in Klein's example the barrier was infinitely broad this could not be associated with wave mechanical tunnel effect. It is truly a paradox: Electrons too slow to surpass the potential,

may still only be partially reflected.... Even for an infinitely high barrier, i.e., $r_2 = 1$ and energies ≈ 1 MeV, (the reflection coefficient) R is less than 75%! From (2) and (3) it appears that as soon as the barrier is sufficiently high: $V > 2mc^2$, electrons may transgress the repulsive wall — seemingly defying conservation of energy.... Nor is it possible by way of the positive energy spectrum of the free electron to achieve complete Einstein causality.”

- ³ Equation (69) is erroneously interpreted as a physical law of the indeterminate nature of conjugate parameters of atomic particles such as position and momentum or energy and time. This so-called Heisenberg uncertainty principle is not a physical law, rather it is a misinterpretation of applying the Schwartz inequality to a probability-wave model of a particle.⁽⁵⁴⁾ The mathematical consequence is that a particle such as an electron can have a continuum of momenta and positions with a continuum of energies simultaneously, which cannot be physical. This result is independent of error or limitations introduced by measurement. Jean B. Fourier was the first to discover the relationship between time and frequency compositions of physical measurables. Equation (69) expresses the limitation of measuring these quantities since an impulse contains an infinity of frequencies, and no instrument has such bandwidth. Similarly, an exact frequency requires an infinite measurement time, and all measurements must be finite in length. Thus (69) is a statement about the limitations of measurement in time and frequency. It is further a conservation statement of energy of a signal in the time and frequency domains. Werner Heisenberg's substitution of momentum and position for a single-particle probability wave into this relationship says nothing

about conjugate parameters of a particle in the absence of their measurement or the validity of the probability-wave model. In fact, this approach has been shown to be flawed experimentally in the Wave-Particle Duality section and Appendix II: Quantum Electrodynamics (QED) is Purely Mathematical and Has No Basis in Reality of Ref. 7, and discussed previously.^(2,9,10)

References

1. F. Laloë, *Am. J. Phys.* **69**, 655 (2001).
2. R.L. Mills, *Ann. Fond. Louis de Broglie* **30**, 877 (2005).
3. *Idem*, *Phys. Essays* **16**, 433 (2003).
4. *Idem*, *Phys. Essays* **18**, 321 (2005).
5. *Idem*, *Phys. Essays* **17**, 342 (2004).
6. *Idem*, "Exact Classical Quantum Mechanical Solution for Atomic Helium Which Predicts Conjugate Parameters from a Unique Solution for the First Time," submitted; posted at http://www.blacklightpower.com/theory/theorypapers/Exact_Solutions_Atomic_Helium_102804.pdf
7. *Idem*, *The Grand Unified Theory of Classical Quantum Mechanics*, October 2007 edition; posted at <http://www.blacklightpower.com/theory/book.shtml>
8. *Idem*, *Int. J. Hydrogen Energy* **27**, 565 (2002).
9. *Ibid.* **26**, 1059 (2001).
10. *Ibid.* **25**, 1171 (2000).
11. H. Margenau and G.M. Murphy, *The Mathematics of Chemistry and Physics* (D. Van Nostrand Company, New York, 1943), pp. 77–78.
12. D.A. McQuarrie, *Quantum Chemistry* (University Science Books, Mill Valley, CA, 1983), p. 291.
13. H. Margenau and G.M. Murphy, *The Mathematics of Chemistry and Physics*, 2nd edition (Krieger Publishing, Huntington, NY, 1976).
14. H.J. Maris, *J. Low Temp. Phys.* **120**, 173 (2000).
15. C.A. Fuchs and A. Peres, *Physics Today* **53** (March 2000), p. 70.
16. S. Peil and G. Gabrielse, *Phys. Rev. Lett.* **83**, 1287 (1999).
17. F.J. Dyson, *Am. J. Phys.* **58**, 209 (1990).
18. J. Horgan, *Sci. Am.* **267**, 94 (1992).
19. V.F. Weisskopf, *Rev. Modern Phys.* **21**, 305 (1949).
20. A. Beiser, *Concepts of Modern Physics*, 4th Edition (McGraw-Hill, New York, 1978), pp. 119–122.
21. M.M. Waldrop, *Science* **242**, 1248 (1988).
22. A. Einstein, B. Podolsky, and N. Rosen, *Phys. Rev.* **47**, 777 (1935).
23. H. Wergeland, in *Old and New Questions in Physics, Cosmology, Philosophy, and Theoretical Biology*, edited by A. van der Merwe (Plenum Press, New York, 1983), pp. 503–515.
24. W.E. Lamb and R.C. Retherford, *Phys. Rev.* **72**, 241 (1947).
25. H.A. Bethe, P.A.M. Dirac, W. Heisenberg, and E.P. Wigner, *From a Life of Physics*, edited by A. Salam et al. (World Scientific, Singapore, 1989).
26. P.W. Milonni, *The Quantum Vacuum: An Introduction to Quantum Electrodynamics* (Academic Press, Boston, 1994), p. 90.
27. P.A.M. Dirac, *Directions in Physics*, edited by H. Hora and J. R. Shepanski (Wiley, New York, 1978), p. 36.
28. H. Dehmelt, *Science* **247**, 539 (1990).
29. H.A. Bethe, *Phys. Rev.* **72**, 339 (1947).
30. L. de Broglie, in *Old and New Questions in Physics, Cosmology, Philosophy, and Theoretical Biology*, edited by A. van der Merwe (Plenum Press, New York, 1983), pp. 83–86.
31. D.C. Cassidy, *Uncertainty: the Life and Science of Werner Heisenberg* (W.H. Freeman and Company, New York, 1992), pp. 224–225.
32. P.J. Mohr and B.N. Taylor, *Rev. Modern Phys.* **72**, 376, 474 (2000).
33. P.W. Milonni, *The Quantum Vacuum: An Introduction to Quantum Electrodynamics* (Academic Press, Boston, 1994), pp. 107–111.
34. *Ibid.*, p. 108.
35. G.P. Lepage, in *Proceedings of the International Conference on Atomic Physics*, Vol. 7 (World Scientific, Singapore, 1981), pp. 297–311.
36. G. Landvogt, *Internat. J. Hydrogen Energy* **28**, 1155 (2003).
37. P. Pearle, *Found. Phys.* **7**, 931 (1977).
38. F. Dyson, *Am. J. Phys.* **58**, 209 (1990).
39. H.A. Haus, *Am. J. Phys.* **54**, 1126 (1986).
40. <http://www.blacklightpower.com/theory/Spreadsheets/1-20%20Electron%20Atoms%20Spreadsheets%20Unprotected.xls>
41. Ref. 12, pp. 206–225.
42. J. Daboul and J.H.D. Jensen, *Z. Phys.* **265**, 455 (1973).
43. T.A. Abbott and D.J. Griffiths, *Am. J. Phys.* **53**, 1203 (1985).
44. G. Goedecke, *Phys. Rev. B* **135**, 281 (1964).
45. Ref. 12, pp. 238–241.
46. R.C. Weast, *CRC Handbook of Chemistry and Physics*, 68th edition (CRC Press, Boca Raton, FL, 1987–88), pp. F-186–F-187.
47. R.S. Van Dyck Jr., P. Schwinberg, and H. Dehmelt, *Phys. Rev. Lett.* **59**, 26 (1987).

48. E.R. Williams and P.T. Olsen, Phys. Rev. Lett. **42**, 1575 (1979).
49. K. v. Klitzing, G. Dorda, and M. Pepper, Phys. Rev. Lett. **45**, 494 (1980).
50. C.E. Moore, "Ionization Potentials and Ionization Limits Derived from the Analyses of Optical Spectra," Nat. Stand. Reference Data Ser.-Nat. Bur. Stand. (U.S.), No. 34, 1970.
51. R.C. Weast, *CRC Handbook of Chemistry and Physics*, 58th edition (CRC Press, West Palm Beach, FL, 1977), p. E-68.
52. J.D. Jackson, *Classical Electrodynamics*, 2nd edition (John Wiley and Sons, New York, 1975), pp. 758–763.
53. W. McC. Siebert, *Circuits, Signals, and Systems* (The MIT Press, Cambridge, MA, 1986), pp. 488–502.
54. Ref. 12, pp. 135–140.
55. T.C. Gibb, *Principles of Mössbauer Spectroscopy* (Chapman and Hall, London, 1977), Chap. 1.
56. U. Gonser, *Mössbauer Spectroscopy* (Springer-Verlag, New York, 1975), pp. 1–51.
57. Ref. 52, pp. 739–779.
58. *Ibid.*, pp. 236–240.
59. L.C. Shen and J.A. Kong, *Applied Electromagnetism* (Brooks/Cole Engineering Division, Monterey, CA, 1983), pp. 170–209.
60. Ref. 12, pp. 27–28.
61. P.J. Mohr and B.N. Taylor, Rev. Modern Phys. **72**, 371 (2000).
62. Ref. 52, pp. 503–561.
63. P.J. Mohr and B.N. Taylor, Rev. Modern Phys. **72**, 418–419 (2000).
64. Ref. 52, pp. 236–240, 601–608, 786–790.
65. *Ibid.*, p.178.
66. P.J. Mohr and B.N. Taylor, Reviews of Modern Physics **72**, 418 (2000).
67. S. Patz, Cardiovasc. Interven. Radiol. **8**(25), 225 (1986).
68. W. Liu et al., Phys. Rev. Lett. **82**, 711 (1999).
69. P.J. Mohr and B.N. Taylor, Rev. Modern Phys. **72**, 448–453 (2000).
70. *Idem*, "CODATA recommended values of the fundamental physical constants: 2002," in press, <http://physics.nist.gov/constants>
71. M.W. Ritter, P.O. Egan, V.W. Hughes, and K.A. Woodle, Phys. Rev. A **30**, 1331 (1984).
72. Ref. 52, p. 9.

Randell L. Mills

BlackLight Power, Inc.
493 Old Trenton Road
Cranbury, New Jersey 08512 U.S.A.

e-mail: rmills@blacklightpower.com